

PERIODIC TABLE

Atomic Properties of the Elements

NIST
National Institute of
Standards and Technology
U.S. Department of Commerce

Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data														
For the most accurate values of these and other constants, visit physics.nist.gov/constants . 1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³ Cs.																		physics.nist.gov		www.nist.gov/srd														
speed of light in vacuum <i>c</i> 299 792 458 m s ⁻¹ (exact) Planck constant <i>h</i> 6.626 069 3 × 10 ⁻³⁴ J s (h = 6.596) elementary charge <i>e</i> 1.602 176 634 × 10 ⁻¹⁹ C electron mass <i>m_e</i> 9.109 382 15 × 10 ⁻³¹ kg proton mass <i>m_p</i> 1.672 621 63 × 10 ⁻²⁷ kg fine-structure constant <i>α</i> 1/137.035 999 074 Rydberg constant <i>R_∞</i> 10 973 731.568 16 × 10 ¹⁰ m ⁻¹ <i>R_H</i> 1.096 775 8 × 10 ⁷ m ⁻¹ <i>R_∞</i> 1.097 373 156 85 × 10 ⁷ m ⁻¹ Boltzmann constant <i>k</i> 1.380 650 5 × 10 ⁻²³ J K ⁻¹																		Solids		Liquids		Gases		Artificially Prepared										
																		13 IIIA		14 IVA		15 VA		16 VIA		17 VIIA		18 VIIIA						
1	H Hydrogen (1.00794) 1.008																	5	B Boron 10.811 10.811 10.811	6	C Carbon 12.011 12.011 12.011	7	N Nitrogen 14.007 14.007 14.007	8	O Oxygen 15.999 15.999 15.999	9	F Fluorine 18.998 18.998 18.998	10	Ne Neon 20.180 20.180 20.180					
2	Li Lithium 6.941 6.941 6.941	Be Beryllium 9.012 9.012 9.012																	13	Al Aluminum 26.982 26.982 26.982	14	Si Silicon 28.086 28.086 28.086	15	P Phosphorus 30.974 30.974 30.974	16	S Sulfur 32.065 32.065 32.065	17	Cl Chlorine 35.453 35.453 35.453	18	Ar Argon 39.948 39.948 39.948				
3	Na Sodium 22.990 22.990 22.990	Mg Magnesium 24.305 24.305 24.305	3	IIIB	4	IVB	5	VB	6	VIB	7	VIIB	8	VIII	9	VIII	10	VIII	11	IB	12	IIB												
4	K Potassium 39.098 39.098 39.098	Ca Calcium 40.078 40.078 40.078	Sc Scandium 44.956 44.956 44.956	Ti Titanium 47.88 47.88 47.88	V Vanadium 50.942 50.942 50.942	Cr Chromium 51.996 51.996 51.996	Mn Manganese 54.938 54.938 54.938	Fe Iron 55.845 55.845 55.845	Co Cobalt 58.933 58.933 58.933	Ni Nickel 58.693 58.693 58.693	Cu Copper 63.546 63.546 63.546	Zn Zinc 65.38 65.38 65.38	Ga Gallium 69.723 69.723 69.723	Ge Germanium 72.63 72.63 72.63	As Arsenic 74.922 74.922 74.922	Se Selenium 78.96 78.96 78.96	Br Bromine 79.904 79.904 79.904	Kr Krypton 83.798 83.798 83.798																
5	Rb Rubidium 85.468 85.468 85.468	Sr Strontium 87.62 87.62 87.62	Y Yttrium 88.906 88.906 88.906	Zr Zirconium 91.224 91.224 91.224	Nb Niobium 92.906 92.906 92.906	Mo Molybdenum 95.94 95.94 95.94	Tc Technetium (98) 98 98 98	Ru Ruthenium 101.07 101.07 101.07	Rh Rhodium 102.906 102.906 102.906	Pd Palladium 106.42 106.42 106.42	Ag Silver 107.868 107.868 107.868	Cd Cadmium 112.411 112.411 112.411	In Indium 114.818 114.818 114.818	Sn Tin 118.710 118.710 118.710	Sb Antimony 121.757 121.757 121.757	Te Tellurium 127.60 127.60 127.60	I Iodine 126.905 126.905 126.905	Xe Xenon 131.29 131.29 131.29																
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				89	Ac Actinium (227) 227 227	90	Th Thorium 232.038 232.038 232.038	91	Pa Protactinium 231.036 231.036 231.036	92	U Uranium 238.029 238.029 238.029	93	Np Neptunium (237) 237 237	94	Pu Plutonium (244) 244 244	95	Am Americium (243) 243 243	96	Cm Curium (247) 247 247	97	Bk Berkelium (247) 247 247	98	Cf Californium (251) 251 251	99	Es Einsteinium (252) 252 252	100	Fm Fermium (257) 257 257	101	Md Mendelevium (258) 258 258	102	No Nobelium (259) 259 259	103	Lr Lawrencium (262) 262 262	

Atomic Number

Ground-state Ionization Energy (eV)

Symbol

Name

Atomic Weight

Ground-state Configuration

Ionization Energy (eV)

58

Ce

Cerium

140.116

(approx.) 5.5057

Atomic Number
 Symbol
 Name
 Atomic Weight
 Ground-state Configuration
 Ionization Energy (eV)

58
Ce
 Cerium
 140.116
 [Xe]4f¹5d¹6s²
 5.5387

*Based upon ^{12}C . () indicates the mass number of the longest-lived isotope.

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