

**Table 2 – Data Collected for Nitric Acid Synthesis**

	1	2	Temperature (°C)				30	40	50	60	70	80
1	10		Nitric Acid Synthesis				10		20	30	40	
2	20						20		30	40	50	
3	30	40					30		40	50	60	
4	40	50					40		50	60	70	
5	50	60					50		60	70	80	
6	60	70			70			70				80
7	70	80				80						

(a) Describe how you find the selected data represent the overall trend observed for the reaction that is studied in

(b) Compare the trend found in Table 2.

(c) The general rate equation for the reaction is:

Rate =

(d) Determine the rate of reaction for the reaction:

Rate =

(e) Which chemical species is the limiting reagent in the reaction?

Answer:

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