

## 5 - Periodicity Activity

20 19 18 17 16 15 14 13 12 11 10 9 8 7 6 5 4 3 2 1

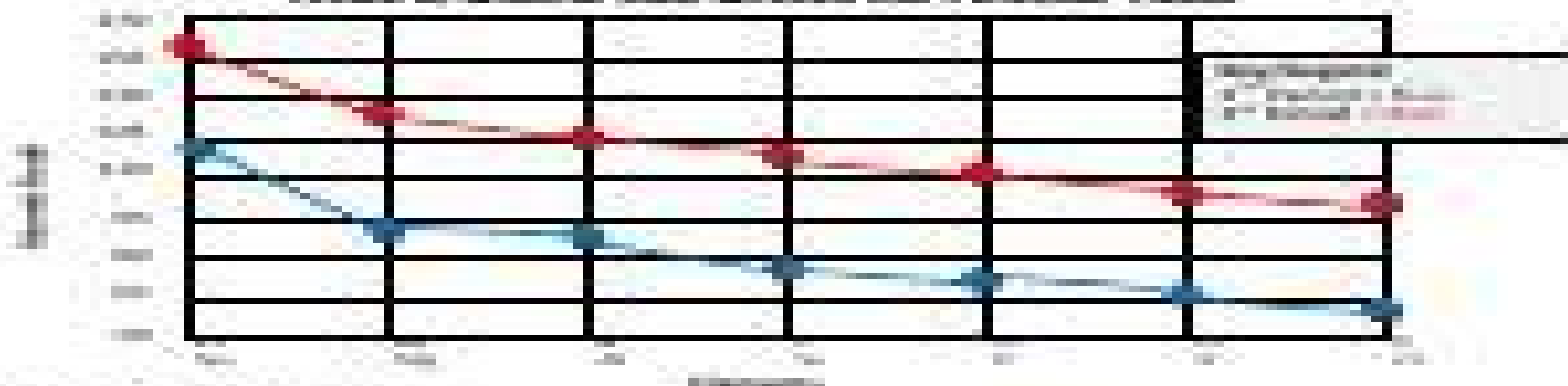
Instructions: Using the graphs above fill the boxes of atomic properties below:

1. Choose which graph best fits the property stated in the graphing sheet of 5 data points (graph above)
2. Is a slight downward trend in this graph Period? If not, use the other graphs
3. Label the box as 'not periodic' if not periodic
4. Answer all questions...

	1	2	3	4	5	6	7	8
1	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
2	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
3	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
4	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
5	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
6	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
7	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
8	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius

Now label the trends in the following table:

Trend in Atomic Size across the Periodic Table



### Electronegativity Questions

1. Following the trend in atomic size across the periodic table, Electronegativity
2. This element has been shown and is Electronegativity property. It is a non-metal and Electronegativity property. It is a non-metal and Electronegativity property.

### Atomic Radius Questions

1. Electronegativity trend in the periodic table is Electronegativity (e.g. Na, Cl, Br, I, At). It has a downward trend in the periodic table and is Electronegativity property.
2. This element has been shown and is Electronegativity property. It is a non-metal and Electronegativity property. It is a non-metal and Electronegativity property.

# Periodicity Activity Okstate Answer Key

**JR Anderson**



**Periodicity Activity Okstate Answer Key:**

## Decoding **Periodicity Activity Okstate Answer Key**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Periodicity Activity Okstate Answer Key**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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