

Periodic Trends: Straw Lab

Objective: To create a visual representation of the following periodic trends: atomic size, ionization energy and electronegativity.

Atomic Radius: the size of an atom measured in either nanometers (nm) or angstroms (Å).

Ionization Energy: the energy needed to remove an electron from an atom measured in kilojoules per mole (kJ/mol) of an atom.

Electronegativity: the relative ability of an atom to attract electrons towards itself.

Materials: For this lab you will need a sheet with 3 small periodic tables, straws, scissors, a metric ruler, your notebook and access to a hot glue gun.

Procedure:

- 1) Draw and complete the data table for each trend in your lab notebook. Each person should have their own data tables in their own notebooks!
- 2) For each trend, calculate the scale factor according to the instructions. Using the scale factor, measure the straw and cut it to the appropriate size.
- 3) You will then use the glue gun to glue the straw to the periodic table in the box that represents the element's location on the periodic table.
- 4) Answer the questions for each trend in your lab notebook. Everyone should have their own answers in their own notebook!

Atomic Radius:

The radius of the following atoms has been estimated to have the following values:

Name	Symbol	Atomic Number	Size of Radius (nm)	Calculated Straw Length (cm)
Fluorine			0.064	
Chlorine			0.0999	
Bromine			0.114	
Iodine			0.1133	
Sodium			0.186	
Magnesium			0.160	
Aluminum			0.143	
Silicon			0.117	

Periodic Trends Straw Lab Answer Key

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Table of Contents Periodic Trends Straw Lab Answer Key

1. Understanding the eBook Periodic Trends Straw Lab Answer Key
 - The Rise of Digital Reading Periodic Trends Straw Lab Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Straw Lab Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends Straw Lab Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Straw Lab Answer Key
 - Personalized Recommendations
 - Periodic Trends Straw Lab Answer Key User Reviews and Ratings
 - Periodic Trends Straw Lab Answer Key and Bestseller Lists
5. Accessing Periodic Trends Straw Lab Answer Key Free and Paid eBooks
 - Periodic Trends Straw Lab Answer Key Public Domain eBooks

- Periodic Trends Straw Lab Answer Key eBook Subscription Services
- Periodic Trends Straw Lab Answer Key Budget-Friendly Options
- 6. Navigating Periodic Trends Straw Lab Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Trends Straw Lab Answer Key Compatibility with Devices
 - Periodic Trends Straw Lab Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Trends Straw Lab Answer Key
 - Highlighting and Note-Taking Periodic Trends Straw Lab Answer Key
 - Interactive Elements Periodic Trends Straw Lab Answer Key
- 8. Staying Engaged with Periodic Trends Straw Lab Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends Straw Lab Answer Key
- 9. Balancing eBooks and Physical Books Periodic Trends Straw Lab Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Trends Straw Lab Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Trends Straw Lab Answer Key
 - Setting Reading Goals Periodic Trends Straw Lab Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Trends Straw Lab Answer Key
 - Fact-Checking eBook Content of Periodic Trends Straw Lab Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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