

Name: \_\_\_\_\_  
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**Exceed to Succeed**

## **Worksheet: Periodic Trends**

### **1. ATOMIC RADIUS**

For each of the following sets of atoms, rank the atoms from smallest to largest atomic radius.

- a. Li, C, F
- b. Li, Na, K
- c. Ge, P, O
- d. C, N, Al
- e. Al, Cl, Ga

### **2. IONIC RADIUS**

For each of the following sets of ions, rank them from smallest to largest ionic radius.

- a.  $\text{Mg}^{2+}$ ,  $\text{Si}^{4+}$ ,  $\text{S}^{2-}$
- b.  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Ba}^{2+}$
- c.  $\text{F}^-$ ,  $\text{Cl}^-$ ,  $\text{Br}^-$
- d.  $\text{Ba}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{Zn}^{2+}$
- e.  $\text{Si}^{4+}$ ,  $\text{P}^{3-}$ ,  $\text{O}^{2-}$

### **3. IONIZATION ENERGY**

For each of the following sets of atoms, rank them from lowest to highest ionization energy.

- a. Mg, Si, S
- b. Mg, Ca, Ba
- c. F, Cl, Br
- d. Ba, Cu, Ne
- e. Si, P, He

### **4. ELECTRONEGATIVITY**

For each of the following sets of atoms, rank them from lowest to highest electronegativity.

- a. Li, C, N
- b. C, O, Ne
- c. Si, P, O
- d. K, Mg, P
- e. S, F, He

# Periodic Trends Chapter 5 Answer Key

**Yicheng Fang**



## **Periodic Trends Chapter 5 Answer Key:**

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Periodic Trends Chapter 5 Answer Key**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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