

Percent, Actual, and Theoretical Yield



a) I began this reaction with 20 grams of lithium hydroxide. What is my theoretical yield of lithium chloride?

b) I actually produced 6 grams of lithium chloride. What is my percent yield?



a) If I start with 5 grams of C_3H_8 , what is my theoretical yield of water?

b) I got a percent yield of 75%. How many grams of water did I make?



My theoretical yield of beryllium chloride was 10.7 grams. If my actual yield was 4.5 grams, what was my percent yield?



What is my theoretical yield of sodium oxide if I start with 20 grams of calcium oxide?

Percent Actual And Theoretical Yield Answer Key

Annelies Wilder-Smith



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Introduction to Chemical Principles Howard Stephen Stoker, 1986 Chemical Principles, Properties, and Reactions

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