#### Topics:

- 1. General Chemistry
- 2. Molecular Representations
- 3. Nomenclature
- 4. Alkanes/Cycloalkanes
- 5. Stereoisomers
- 6. Chemical reactivity/mechanism
- 7. SN/E Reactions

#### General Chemistry Questions

- 1. Identify all of the intermolecular forces present in each molecule:
  - a. CH<sub>2</sub>OH
  - b. CH<sub>2</sub>COCH<sub>2</sub>CH<sub>3</sub>
  - c. CH<sub>2</sub>C(CH<sub>3</sub>)<sub>2</sub>
- 2. Fill out the blanks in the chart (VESPR Theory)

Hybridization	Geometry	# of Bonds	# of Lone Pairs	Bond Angle
	Tetrahedral	The state of the s	0	The second second
Sp <sup>3</sup>		3	1	
	Bent			109.5
Sp²	3,550	3	0	
		2	1	120
			0	180

- 3. Determine if the following molecules have a dipole moment
  - a. CH<sub>2</sub>CH<sub>3</sub>
  - b. CH3CH2CH(NH<sub>2</sub>)CH<sub>3</sub>
  - c. CH<sub>2</sub>Cl<sub>2</sub>
  - d. H<sub>2</sub>O\*
  - e. HOOst

#### Molecular Representations Questions:

1. Write the condensed name for the following structures:



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SJ Ball

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