Saturated and Unsaturated Solutions

Is there a limit to the amount of solute that will dissolve in a solvent?

Why?

We use solutions every day. People who wear contact lenses use "lens solution" to rinse their contacts and keep them wet. Athletes who consume sports drinks after exercising benefit from the electrolytes in those solutions. This activity will explore whether or not there is a limit to how much of one substance can dissolve in another.

Model 1 - Saturated and Unsaturated Solutions

	Unsaturated Solutions	
All beakers contain 10.0 g of water. All beakers are kept at 20 °C. All solutions are stirred for 2 hours. Solute is the same substance in all beakers.	Beaker A o o o 1.0 g of solute added	Beaker B
Number of dissolved particles	5	
Number of solid particles		0

	Saturated Solutions	
Beaker C	Beaker D	Beaker E
		0000000
3.6 g of solute added	7.0 g of solute added	9.0 g of solute added
-	17	

Rules Of Solubility Pogil Answers

TD Snyder

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Solubility Rules Based on Lattice Parameter Differences U. Bendict, 1977 **Solubilities of Inorganic and** Organic Substances ,1907 Regular and Related Solutions Joel Henry Hildebrand, 2000 Solubility of Solids in Solubilities of Inorganic and Organic Substances ,1907 Dense Gases J. M. Prausnitz, 1965 **Solubilities of Inorganic and Organic Compounds** Atherton Seidell, 1928 Principles of Solution and Solubility Kozo Shinoda, 1991 Solubilities of Inorganic and Organic Substances Atherton Seidell, 1919 **Solubilities of Inorganic and Organic Compounds** Atherton Seidell, 1928 The Solubility of Nonelectrolytes Joel Henry Hildebrand, Robert Lane Scott, 1964 Solubility data series International Union of Pure and Applied Chemistry. Commission on Solubility Data, 1979 Solubility Data Series International Union of Pure and Applied Solutions and Solubilities Michael R. J. Dack, 1976 Chemistry. Commission of Equilibrium Data. Sub-Committee on Solubility Data, 1979 Solubilities of inorganic and organic compounds Atherton Seidell, William F. Linke, 1952 First Outlines of a Dictionary of Solubilities of Chemical Substances Frank Humphreys Storer, 1863 Solubility Data Series International Union of Pure and Applied Chemistry, 1979 Solubility in Inorganic Two-component Systems Miroslav Broul, 1981

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