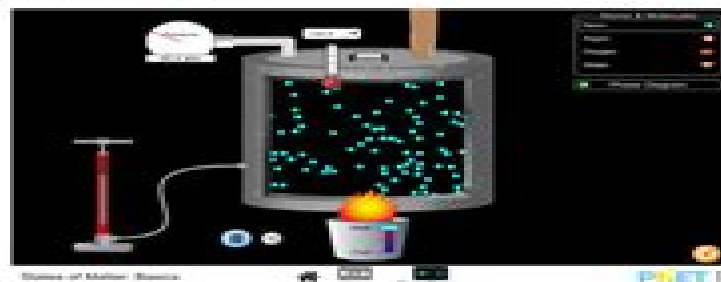


Gas Laws Simulation Lab

- Go to [Phet.colorado.edu](http://phet.colorado.edu)
- Search for the "States of Matter: Basics" simulation.
- Click on the "Phase Changes" icon.
- Select "Oxygen."



1. The simulation shows a quantity of oxygen molecules. What state of matter is this?
2. How do you know?
3. Heat the oxygen to about 150 K. What state of matter is this?
4. How do you know?

Experiment One: Constant volume.

- Leave the oxygen at 150 K. In the table below, record the pressure and temperature of the oxygen in the first row.
- Heat the oxygen to about 450 K.
- Record the new temperature and pressure in the second row of the table.
- Heat the oxygen to about 900 K.
- Record the new temperature and pressure in the third row of the table.

Temperature vs. pressure at constant volume	
Temperature (K)	Pressure (atm)

5. Write a statement describing how the temperature of a quantity of a gas is related to its pressure at constant volume.

Phet Lab Gas Variables And Laws Answers

Sir James Augustus Henry Murray



Phet Lab Gas Variables And Laws Answers:

The Compact Edition of the Oxford English Dictionary Sir James Augustus Henry Murray, 1971 Micrographic reproduction of the 13 volume Oxford English dictionary published in 1933

Phet Lab Gas Variables And Laws Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Phet Lab Gas Variables And Laws Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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