

Click on the **Width** icon to open the measuring tool. Adjust the **temperature** to 300 K. In the "**Hold Constant**" box, select **Pressure (P)** to keep the pressure constant.

6) What is the current **width** of the container and what **units** are used?

10.0 mm

Complete the chart below by changing the **volume** (width) and recording the **temperature**.



Volume Size	Width (mm)	Temperature (K)
Extra Large	14.0	~ 420
Large	12.0	~ 360
Medium	10.0	~ 300
Small	8.0	~ 240
Extra Small	6.0	~ 180



7) Using the data in the chart, describe the relationship between **temperature** and **volume**:

*As the volume decreases, the temperature decreases.
As the volume increases, the temperature increases.*

- A **direct relationship** describes a relationship where two variables rise and fall together.
- An **inverse relationship** describes a relationship where two variables change in opposite directions. If one variable rises, the other variable falls.

8) What type of relationship is present in **Boyle's Law**?

Inverse Relationship

9) What type of relationship is present in **Charles' Law**?

Direct Relationship

Phet Gas Law Simulation Lab Answers

Tao Wei



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