

Name: _____
 Hour: _____ Date: _____

Chemistry: pH and pOH calculations

Part 1: Fill in the missing information in the table below.

pH	[H_3O^{1+}]	pOH	[OH^{2-}]	ACID or BASE?
3.78				
	$3.89 \times 10^{-4} \text{ M}$			
		5.19		
			$4.88 \times 10^{-6} \text{ M}$	
8.46				
	$8.43 \times 10^{-13} \text{ M}$			
		2.14		
			$2.31 \times 10^{-11} \text{ M}$	
10.91				
	$7.49 \times 10^{-6} \text{ M}$			
		9.94		
			$2.57 \times 10^{-8} \text{ M}$	
4.16				
	$1.06 \times 10^{-3} \text{ M}$			
		3.82		
			$8.53 \times 10^{-7} \text{ M}$	
7.05				
	$4.73 \times 10^{-10} \text{ M}$			
		1.33		
			$9.87 \times 10^{-3} \text{ M}$	
11.68				
	$9.22 \times 10^{-8} \text{ M}$			
		12.24		
			$5.39 \times 10^{-12} \text{ M}$	

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Clemens Wendtner



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