

Gas Laws Graphing Activity

Students will do the following objectives:

1. Demonstrate understanding of Boyle's, Charles', and Gay-Lussac's laws and their applications.
2. Explain the behavior of gases.
3. Demonstrate the relationship between volume and pressure.
4. Demonstrate the relationship between volume and temperature.
5. Explain the meaning of Boyle's, Charles', and Gay-Lussac's laws.
6. Describe a gas mixture.

Procedure:

1. Prepare a document in your computer. Insert each part of the activity into the document in the appropriate order.

2. Graph volume and pressure, volume and temperature, and graph of the relationship between them.

3. Explain the volume-pressure relationship and Charles' law, and explain both pressure and volume relationships with the relationship between them.

4. Save the document with appropriate file name.

Part 1: Pressure and Temperature

1. Prepare the document with Boyle's, Charles', and Gay-Lussac's laws. Make a separate document after.

Note:

Pressure (atm)	Temperature (K)
1.00	273
1.00	273
1.00	273
1.00	273
1.00	273
1.00	273

Graph:

Graph Name:

Relationship between volume and pressure, volume and temperature, and volume and temperature relationships.

Boyle's Law:

Relationship between volume and temperature.

Relationship between volume and temperature.

Phet Gas Laws Graphing Activity

Rosina Ehmman



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