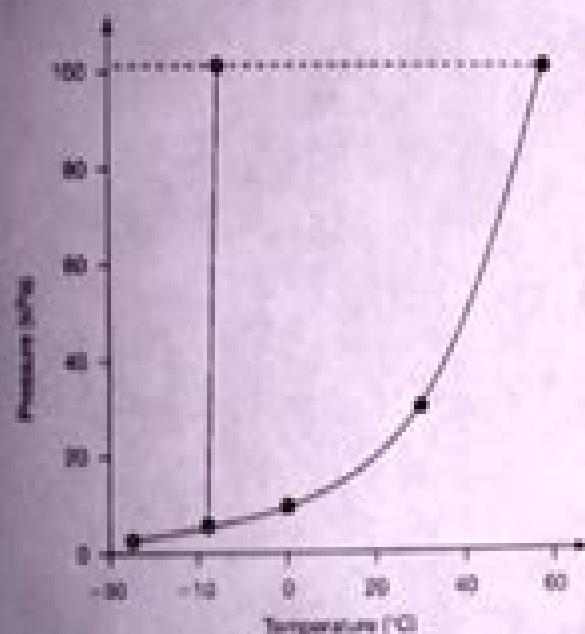


4. The image below shows the phase diagram of bromine, which exists as a red liquid at room temperature and pressure. Use this phase diagram to answer the questions (letters a through e).



a) Label the different regions of this phase diagram as solid, liquid and gas.

b) What is the normal boiling point of this substance?

c) Indicate the conditions of the triple point of this substance.

d) What is the phase of this substance at a pressure of 20 kPa and 10°C?

e) If a sample of bromine is at an initial pressure of 95 kPa and an initial temperature of -20°C, what phase change(s) will occur as the sample is warmed to 60°C while maintaining the pressure? Indicate the temperature value for any phase changes.

# Phase Diagram Of Bromine Answer Key

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## Phase Diagram Of Bromine Answer Key:

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