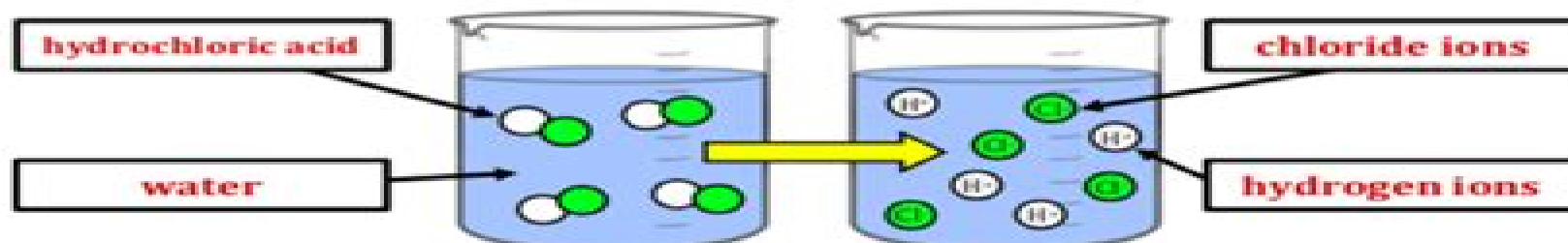


Introduction to Acids and Bases

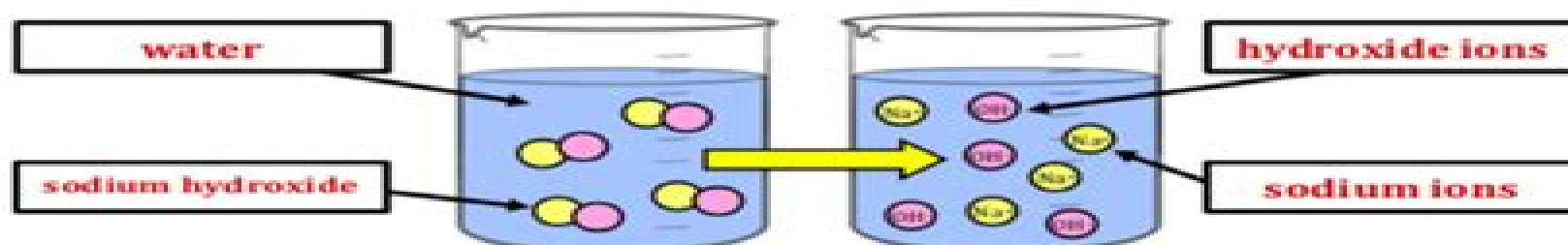
According to the **Arrhenius** definition:

- **Acids** are substances that release **hydrogen** ions in water (H^+)
- **Bases** are substances that release **hydroxide** ions in water (OH^-)

Instructions: The diagram below shows hydrochloric acid dissolving in water.
Label water, hydrochloric acid, hydrogen ions, and chloride ions.



Instructions: The diagram below shows sodium hydroxide dissolving in water.
Label water, sodium hydroxide, sodium ions, and hydroxide ions.



Instructions: Identify whether the chemicals below would be *acids* or *bases*.

- | | |
|----------------------------|-------------|
| 1) HBr | <u>acid</u> |
| 2) LiOH | <u>base</u> |
| 3) H_2SO_4 | <u>acid</u> |

- | | |
|-----------------------------|-------------|
| 4) $\text{Ca}(\text{OH})_2$ | <u>base</u> |
| 5) HNO_3 | <u>acid</u> |
| 6) KOH | <u>base</u> |

Note Taking Worksheet Solutions Acids And Bases

**Mary Jones, Richard Harwood, Ian
Lodge, David Sang**



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vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students

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