

Atomic Number

- Increase across a period
- increase down a family

Atomic Mass

- increase across a period
- increase down a family

Atomic Radius

half the distance between nuclei of two like atoms.

- decreases across a period (due to increased attraction of more positive nuclei to electron clouds)
- increase down a family (due to addition of an energy level)

Electron Affinity

amount of energy when an atom gains an electron

- increases across a period
- decreases down a family

Ionization Energy

amount of energy required to remove an electron from a neutral atom

- increase across a period
- decrease down a family

Electron Distribution - Forming Ions

- Atoms become ions by gaining or losing electrons to become an ion.
- An atom is stable when it has a full outer energy level (8 valence electrons).
- Atoms that lose (give away) electrons become positive charged ions. Atoms that gain (steal) electrons become negative charged ions.

Note Taking Guide Episode 403

Yi-Tong Ma

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name Yi-Tong Ma.

Note Taking Guide Episode 403:

Note Taking Guide Episode 403 Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Note Taking Guide Episode 403**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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