



W1-2-60-1-6

**JOMO KENYATTA UNIVERSITY OF AGRICULTURE AND TECHNOLOGY
UNIVERSITY EXAMINATIONS 2016/2017**

**THIRD YEAR SECOND SEMESTER EXAMINATION FOR THE DEGREE OF
BACHELOR OF SCIENCE IN INFORMATION TECHNOLOGY**

STA 2200: PROBABILITY AND STATISTICS II

DATE: APRIL 2017

TIME: 2 HOURS

INSTRUCTIONS:

**ANSWER QUESTION ONE [COMPULSORY] AND ANY OTHER TWO
QUESTIONS**

QUESTION ONE

- (a) State two conditions for a continuous real valued function $f(x)$ to be a probability density function [2 marks]
(b) Let the random variable x have the following probability density function

x	-1	0	1
$F(x)$	$\frac{1}{4}$	$\frac{1}{6}$	$\frac{7}{12}$

Let $E[W(2)]$ denote the expected value of $w(x)$. Find

- (i) $E(x)$ [2 marks]
(ii) $E[2x^2+x+1]$ [4 marks]
(c) Let X be a random variable whose pdf is

$$f(x) = \begin{cases} \frac{e^{-1} x^x}{x^1} & x = 0, 1, 2, \dots \\ 0 & \text{otherwise} \end{cases}$$

- (i) Show that its moment generating function is given by $e^{2(e^x-1)}$ [3 marks]
(ii) Hence or otherwise obtain the mean and variance of X [5 marks]

(d) Let $f(x)$ be the pdf of a random variable x

$$f(x) = \begin{cases} k(1-x)^2 & 0 < x < 1 \\ 0 & \text{otherwise} \end{cases}$$

Probability And Statistid 2 Jkuat Notes

**Christopher John Wild, George Arthur
Frederick Seber**



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web understand that parts of an acid base titration be able to determine the K_a or K_b from ph data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web 1 approximately 3 g of hydrated oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ was weighted to the nearest 0.01 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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