

Polyatomic Ions WS

Ammonium (NH_4)¹⁺
 Carbonate (CO_3)²⁻
 Chromate (CrO_4)²⁻

Dichromate (Cr_2O_7)²⁻
 Hydroxide (OH)¹⁻
 Nitrate (NO_3)¹⁻

Nitrite (NO_2)¹⁻
 Sulfate (SO_4)²⁻
 Sulfite (SO_3)²⁻

Phosphate (PO_4)³⁻

Write the NAME of each of the following compounds.

1. $(\text{NH}_4)\text{Cl}$ Ammonium Chloride
2. $\text{Be}(\text{SO}_4)$ Beryllium Sulfate
3. $(\text{NH}_4)_3\text{N}$ Ammonium Nitride
4. MgCl_2 Magnesium Chloride
5. $\text{NH}_4(\text{NO}_3)$ Ammonium Nitrate
6. $\text{Sr}_3(\text{PO}_4)_2$ Strontium Phosphate
7. $\text{Zn}(\text{CrO}_4)_2$ Zinc Dichromate
8. $\text{K}_2(\text{Cr}_2\text{O}_7)$ Potassium Dichromate

9. $\text{Ga}(\text{ClO}_3)_3$ Gallium Chlorate
10. $\text{Cu}(\text{OH})$ Copper I Hydroxide
11. $(\text{NH}_4)_3(\text{PO}_4)$ Ammonium Phosphate
12. $\text{Fe}(\text{SO}_4)$ Iron II Sulfate
13. $\text{Mg}(\text{NO}_2)_2$ Magnesium Nitrite
14. $(\text{NH}_4)\text{NO}_2$ Ammonium Nitrite
15. $\text{Na}_2(\text{Cr}_2\text{O}_7)$ Sodium Dichromate
16. $\text{Na}(\text{OH})$ Sodium Hydroxide

Write the CHEMICAL FORMULA for each of the given NAMES.....cross charges

- | | | | | |
|-------------------------|-------------------------------------|---|-------------------------------|-------------------|
| 17. calcium carbonate | $\text{Ca}^{2+} (\text{CO}_3)^{2-}$ | = | $\text{Ca}_2(\text{CO}_3)_2$ | = CaCO_3 |
| 18. barium nitrate | $\text{Ba}^{+2} \text{NO}_3^{-1}$ | = | $\text{Ba}(\text{NO}_3)$ | |
| 19. ammonium sulfate | $\text{NH}_4^{+1} \text{SO}_4^{-2}$ | = | $(\text{NH}_4)_2 \text{SO}_4$ | |
| 20. aluminum hydroxide | $\text{Al}^{+3} \text{OH}^{-1}$ | = | $\text{Al}(\text{OH})_3$ | |
| 21. calcium phosphate | $\text{Ca}^{+2} \text{PO}_4^{-3}$ | = | $\text{Ca}_3(\text{PO}_4)_2$ | |
| 22. cesium nitrate | $\text{Cs} \text{NO}_2^{-1}$ | = | CsNO_3 | |
| 23. sodium nitrite | $\text{Na}^{+1} \text{NO}_3^{-1}$ | = | NaNO_3 | |
| 24. calcium sulfate | $\text{Ca}^{+2} \text{SO}_4^{-2}$ | = | CaSO_4 | |
| 25. beryllium sulfate | $\text{Be} \text{SO}_4^{-2}$ | = | BeSO_4 | |
| 26. sodium carbonate | $\text{Na}^{+1} \text{CO}_3^{-2}$ | = | Na_2CO_3 | |
| 27. magnesium phosphate | $\text{Mg}^{+2} \text{PO}_4^{-3}$ | = | $\text{Mg}_3(\text{PO}_4)_2$ | |
| 28. calcium phosphate | $\text{Ca}^{+2} \text{PO}_4^{-3}$ | = | $\text{Ca}_3(\text{PO}_4)_2$ | |

Polyatomic Ions Ws

Arturo Cuomo



Polyatomic Ions Ws:

Unveiling the Energy of Verbal Beauty: An Psychological Sojourn through **Polyatomic Ions Ws**

In a world inundated with displays and the cacophony of immediate connection, the profound energy and emotional resonance of verbal beauty often diminish in to obscurity, eclipsed by the continuous barrage of sound and distractions. However, nestled within the musical pages of **Polyatomic Ions Ws**, a captivating perform of literary splendor that pulses with organic feelings, lies an unforgettable journey waiting to be embarked upon. Composed by a virtuoso wordsmith, this exciting opus guides readers on a psychological odyssey, gently exposing the latent possible and profound influence embedded within the delicate web of language. Within the heart-wrenching expanse of this evocative examination, we can embark upon an introspective exploration of the book is main themes, dissect its fascinating publishing type, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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