

Chapter – 8

B. Very Short Answer Type Questions:

1. Write the molecular formula of aluminium oxide.

Ans: Al_2O_3

2. The elements present in sodium bicarbonate are sodium, carbon, _____ and oxygen.

Ans: Hydrogen

3. Which one has more mass 1 mole of hydrogen gas or 0.20 mole of nitrogen gas?

Ans: 0.20 mole of nitrogen gas

4. Which one has more number of molecules 1 mole of N_2 or 1 mole of O_2 ?

Ans: Same. Because 1 mole of any substance always contain 6.022×10^{23} of that substance.

5. What is the value of Avogadro's constant?

Ans: 6.022×10^{23}

6. What is the valency of carbon in (a) CCl_4 and (b) CO_2 ?

Ans: (a) CCl_4 – 4, (b) CO_2 – 4

7. If the atomic symbol of Cobalt is Co, then what is the atomic symbol of Copper?

Ans: Cu

8. What is the number of moles in 18g of water?

Ans: 1

9. What is the mass of 1 mole of water?

Ans: 18g

C. Short Answer Questions Type - I:

1. State the Law of constant proportion and name the scientist who proposed this law?

Ans: Law of constant proportion states that,

A chemical compound always contains the same element combined together in the same definite ratio by mass.

This law was proposed by Joseph L. Proust.

Sciencechapter 9 Review And Assessment Answer

F Rizvi



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In a digital era where connections and knowledge reign supreme, the enchanting power of language has be more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Sciencechapter 9 Review And Assessment Answer**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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