

Name: _____

Hour: _____

Temperature and Heat

- 2) _____ The flow of energy from an object at a higher temperature to an object at a lower temperature
- 3) _____ The total random kinetic energy of particles in an object
- 4) _____ The amount of energy needed to raise the temperature of one gram of water by 1°C
- 4) _____ The standard scientific unit in which energy is measured
- 5) _____ The metric temperature scale
- 6) _____ The SI unit for temperature
- 7) _____ The amount of energy required to raise the temperature of one gram of a substance by 1°C
- 8) _____ The process that moves energy from one object to another when they are touching physically
- 9) _____ Materials that transfer energy easily
- 10) _____ Are poor conductors that have a high specific heat
- 11) _____ All of the particles that make up matter are constantly in motion
- 12) _____ A measure of the average kinetic energy of all the particles in an object
- 13) _____ Temperature is measured in these units
- 14) _____ Measures temperature through the regular variation of some physical property of the material inside the thermometer
- 15) _____ The process that transfers energy by the movement of large numbers of particles in the same direction within a liquid or gas
- 16) _____ Energy that travels as the electromagnetic waves, it includes visible light, microwaves, and infrared light
- 17) _____ Change in volume of an object due to expansion caused by motion of particles as the temperature increases
- 18) _____ The standard temperature scale

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Department of Science and Art**



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