

Name: _____ KEY _____ Period _____ Date: _____

Fill in the blank (2 points each blank)

1. Elements that have the properties of both metal and non-metals are metalloids.
2. What do elements in the same group have the same number of? Valence electrons
3. Which element would have the highest electron affinity? Fluorine
4. What element is at period 3, group 8A? Argon
5. How many valence electrons does Bromine have? 7
6. The atomic number tells you how many protons (and electrons if neutral) there are in an atom.
7. Two Characteristics of Non-metals are not good conductors of heat or electricity, not malleable, etc.
8. Two characteristics of Metals are malleable, ductile, good conductors of heat and electricity
9. Which element would have the smallest atomic radius? Helium
10. Which element would have the highest ionization energy? Helium
11. For the questions below and based on the information provided, fill in the missing information. (1 point each)

Name	Symbol	# of protons	# of electrons	# of neutrons	Atomic Number	Mass Number
Potassium	K	19	19	22	19	41
Phosphorous	P	15	15	18	15	33
Beryllium	Be	4	4	4	4	8
Sulfur	S	16	16	20	16	36

12. Draw Bohr Models for (3 points each)

a. Na23

b. N15

13. What is an isotope? (1 point)

An atom with a different number of neutrons

14. How can you tell the identity of any atom? (1 point)

Number of PROTONS

15. Alien Periodic Chart -- Directions: Read the clue and place the alien symbol in the corresponding box!

Periodic Table Test With Answer Key

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Table of Contents Periodic Table Test With Answer Key

1. Understanding the eBook Periodic Table Test With Answer Key
 - The Rise of Digital Reading Periodic Table Test With Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Test With Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Test With Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Test With Answer Key
 - Personalized Recommendations
 - Periodic Table Test With Answer Key User Reviews and Ratings
 - Periodic Table Test With Answer Key and Bestseller Lists

5. Accessing Periodic Table Test With Answer Key Free and Paid eBooks
 - Periodic Table Test With Answer Key Public Domain eBooks
 - Periodic Table Test With Answer Key eBook Subscription Services
 - Periodic Table Test With Answer Key Budget-Friendly Options
6. Navigating Periodic Table Test With Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Test With Answer Key Compatibility with Devices
 - Periodic Table Test With Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Test With Answer Key
 - Highlighting and Note-Taking Periodic Table Test With Answer Key
 - Interactive Elements Periodic Table Test With Answer Key
8. Staying Engaged with Periodic Table Test With Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Test With Answer Key
9. Balancing eBooks and Physical Books Periodic Table Test With Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Test With Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Test With Answer Key
 - Setting Reading Goals Periodic Table Test With Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Test With Answer Key
 - Fact-Checking eBook Content of Periodic Table Test With Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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