

The questions in the test bank cover the concepts from the lessons in Chapter 2. The types of questions include multiple choice, true/false, fill-in-the-blank, and short answer.

Chapter II : Atoms, Molecules, and Ions

- Atomic Structure

- The Iron(II) ion has:
 - 28 electrons and a charge of 3+
 - 24 electrons and a charge of 3-
 - 26 electrons and a charge of 2+
 - 24 electrons and a charge of 2+
 - 26 electrons and a charge of 3+
- Determine the number of *electrons* , *protons* , and *neutrons* of $^{56}_{26}\text{Fe}^{3+}$ and $^{204}_{81}\text{Tl}$?
- A neutral atom of the element potassium $^{39}_{19}\text{K}$ has :
 - 19 e, 19 p, 21 n
 - 40 e, 40 p, 19 n
 - 19 e, 21 p, 19 n
 - 40 e, 19 p, 21 n
 - 21 e, 19 p, 19 n
- The number of *neutrons* in an atom of $^{127}_{53}\text{I}$ is :
 - 53
 - 74
 - 127
 - 180
- An atom of the isotope chlorine-39 consists of how many *protons*, *neutrons*, and *electrons*? (p = proton, n= neutron, e = electron)
 - 17 p, 17 n, 22 e
 - 39 p, 38 n, 39 e
 - 22 p, 17 n, 17 e
 - 17 p, 39 n, 17 e
 - 17 p, 22 n, 17 e
- How many **protons** and **electrons** are present in a $^{40}_{20}\text{Ca}^{2+}$ ion?
 - 18 p, 20 e
 - 20 p, 18 e
 - 20 p, 21 e
 - 20 p, 19 e
 - 22 p, 20 e
- The elements in a *row* of the *periodic table* are known as:
 - metalloids
 - noble gases
 - a group
 - nonmetals
 - a period
- True or False :

() The number of electrons in natural atom equal to the number of neutrons

- Isotopes

- Two *isotopes* of an element *differ* in their:
 - atomic mass
 - atomic number
 - number of protons
 - number of electrons
- For the following data about unknown element :

92.21%	(27.977)	4.700%	(28.976)	3.090%	(29.974)
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Identify the element ?

Science Focus Test Chapter Atoms

L Reisser



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