Unit 1 Progress Check: MCQ

- 1. Which of the following numerical expressions gives the number of moles in 5.0 g of CaO?
- (A) 5.0 g × 56 g/mol
- (B) 55 g/mol
- © 56 g/mol
- In a lab a student is given a 21 g sample of pure Cu metal. Which of the following pieces of information is most useful for determining the number of Cu atoms in the sample? Assume that the pressure and temperature in the lab are 1.0 atm and 25°C.
- A The molar mass of Cu
- B The density of Cu at 25 °C
- (C) The volume of the Cu sample
- (D) The ratio of the two main isotopes found in pure Cu
- 3. A 1.0 mol sample of which of the following compounds has the greatest mass?

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Zhe Xu

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Unveiling the Magic of Words: A Report on "Science Lab Progress Check Unit Cquestions"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really aweinspiring. Enter the realm of "Science Lab Progress Check Unit Cquestions," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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