

REVISION: Global and International "Text" in Chemistry in the Environment (Exercises)

Name:	Matr.:	Grp.:	Grp.:	Grp.:
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Write the answers in your own words! (You may use your own words, but you must use the correct scientific terminology!)

Explain the following: Explain the difference between the two types of chemical reactions.

Explain the difference between the two types of chemical reactions.



Part 1: Knowledge and Understanding (40 marks)

Question 1 (10 marks)

What is the difference between the two types of chemical reactions?

1. _____ is a reaction in which the reactants are _____ and the products are _____.
 a) a homogeneous mixture b) a heterogeneous mixture
 c) a mixture of two substances d) a mixture of two substances
2. _____ is a chemical reaction in which the reactants are _____ and the products are _____.
 a) a homogeneous mixture b) a heterogeneous mixture
 c) a mixture of two substances d) a mixture of two substances
3. _____ is a reaction in which the reactants are _____ and the products are _____.
 a) a homogeneous mixture b) a heterogeneous mixture
 c) a mixture of two substances d) a mixture of two substances
4. _____ is a reaction in which the reactants are _____ and the products are _____.
 a) a homogeneous mixture b) a heterogeneous mixture
 c) a mixture of two substances d) a mixture of two substances
5. _____ is a reaction in which the reactants are _____ and the products are _____.
 a) a homogeneous mixture b) a heterogeneous mixture
 c) a mixture of two substances d) a mixture of two substances
6. _____ is a reaction in which the reactants are _____ and the products are _____.
 a) a homogeneous mixture b) a heterogeneous mixture
 c) a mixture of two substances d) a mixture of two substances
7. _____ is a reaction in which the reactants are _____ and the products are _____.
 a) a homogeneous mixture b) a heterogeneous mixture
 c) a mixture of two substances d) a mixture of two substances
8. _____ is a reaction in which the reactants are _____ and the products are _____.
 a) a homogeneous mixture b) a heterogeneous mixture
 c) a mixture of two substances d) a mixture of two substances
9. _____ is a reaction in which the reactants are _____ and the products are _____.
 a) a homogeneous mixture b) a heterogeneous mixture
 c) a mixture of two substances d) a mixture of two substances
10. _____ is a reaction in which the reactants are _____ and the products are _____.
 a) a homogeneous mixture b) a heterogeneous mixture
 c) a mixture of two substances d) a mixture of two substances

What are the products of the reaction?

1. _____ is a reaction in which the reactants are _____ and the products are _____.
 a) a homogeneous mixture b) a heterogeneous mixture
 c) a mixture of two substances d) a mixture of two substances
2. _____ is a reaction in which the reactants are _____ and the products are _____.
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 c) a mixture of two substances d) a mixture of two substances
10. _____ is a reaction in which the reactants are _____ and the products are _____.
 a) a homogeneous mixture b) a heterogeneous mixture
 c) a mixture of two substances d) a mixture of two substances



What is the difference between the two types of chemical reactions?

Sch4c Environmental Chemistry Unit Test

VM Jensen

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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