

**WKS 7.1 – Beginning Naming & Formula Writing (2 pages)**

Name the following ionic compounds. Be sure to check for Roman Numerals!

$\text{Na}_3\text{P}$		$\text{Mg}_2\text{Br}_2$	
$\text{Ag}_2\text{Se}$		$\text{AlCl}_3$	
$\text{Ca}_3\text{P}_2$		$\text{Ba}_2\text{N}_2$	
$\text{Li}_2\text{O}$		$\text{ZnF}_2$	
$\text{K}_2\text{S}$		$\text{SrI}_2$	
$(\text{NH}_4)_2\text{S}$		$\text{Ca}_3(\text{PO}_3)_2$	
$\text{Li}_2\text{CO}_3$		$\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3$	
$\text{KCN}$		$\text{AgNO}_3$	
$\text{Sb}(\text{NO}_3)_3$		$\text{NiS}$	
$\text{Ca}(\text{NO}_3)_2$		$\text{FeCl}_2$	
$\text{Au}(\text{CN})_3$		$\text{Pb}(\text{SO}_4)_2$	

Write the correct formulas for each of the ionic compounds.

silver iodide		lithium sulfide	
magnesium nitride		cadmium bromide	
zinc chloride		barium phosphide	
aluminum fluoride		arcentium oxide	
potassium permanganate		strontium sulfate	
cesium bromate		magnesium cyanide	
aluminum hypiodite		cadmium periodate	
tin (IV) chlorate		bismuth (III) carbonate	
copper (I) sulfide		iron (III) oxide	

# Pre Ap Chemistry Unit 9 Hw Packet Answers

**Mike Jess**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name 'Mike Jess'.

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