

## Colligative Properties Worksheet

*FYI:  $K_f$  of water is  $1.86^\circ \text{C/m}$ ;  $K_b$  of water is  $0.52^\circ \text{C/m}$*

- 1) Explain why the addition of a solute decreases the melting point of a liquid.
- 2) What is the melting point of a solution in which 3.5 grams of sodium chloride is added to 230 mL of water?
- 3) What is the boiling point of a solution in which 22.5 grams of sucrose ( $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ ) is dissolved in 550 mL of water?
- 4) Rank the following solutions from lowest to highest boiling point, a 2.5 m solution of sodium chloride, a 3.5 m solution of magnesium chloride, or a 4.5 m solution of sulfur dioxide?
- 5) If an aqueous solution has a boiling point of  $102.5^\circ$ , what's the effective molality of the solute?

# Practice Test On Colligative Properties

**JS Bruner**



## **Practice Test On Colligative Properties:**

## **Practice Test On Colligative Properties** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Practice Test On Colligative Properties**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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