

AP Chemistry Chapter 20 Electrochemistry Practice Free Response #1



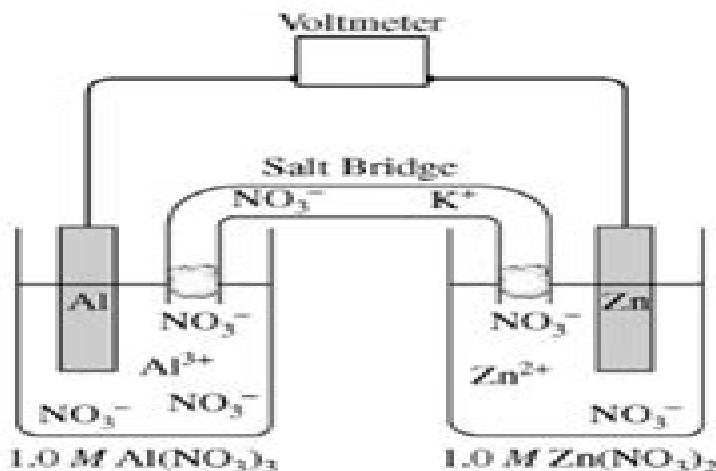
Respond to the following statements and questions that relate to the species and the reaction represented above.

(a) Write the complete electron configuration (e.g.,  $1s^2 2s^2 \dots$ ) for  $\text{Zn}^{2+}$ .

(b) Which species,  $\text{Zn}$  or  $\text{Zn}^{2+}$ , has the greater ionization energy? Justify your answer.

(c) Identify the species that is oxidized in the reaction.

The diagram shows a galvanic cell based on the reaction. Assume that the temperature is  $25^\circ\text{C}$ .



(d) The diagram includes a salt bridge that is filled with a saturated solution of  $\text{KNO}_3$ . Describe what happens in the salt bridge as the cell operates.

(e) Determine the value of the standard voltage,  $E^\circ$ , for the cell.

# Practice Hall Chemistry Chapter 20

**N Noddings**



## **Practice Hall Chemistry Chapter 20:**

The book delves into Practice Hall Chemistry Chapter 20. Practice Hall Chemistry Chapter 20 is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Practice Hall Chemistry Chapter 20, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
    - Chapter 1: Introduction to Practice Hall Chemistry Chapter 20
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