

**Practice Exam AP Chemistry Midterm 1<sup>st</sup> 9 weeks**  
**Section I - Multiple Choice (50%)**

Directions: Each set of lettered choices below refers to the numbered statements immediately following it. Select the one lettered choice that best fits each statement and then blacken the corresponding space on the Scantron. A choice may be once, more than once, or not at all in each set. You must choose 40 out of 50, leave 10 blank. Wrong answers have a 0.25 point penalty, so no guessing!

Questions 1-3

- A. phosphorus
- B. beryllium
- C. sodium
- D. magnesium
- E. sulfur

1. Which of the above is a chalcogen?
2. Which of the above has a nonmetal with an oxidation number of +6?
3. Which of the above forms salts that are all soluble in water?



Questions 4-6

- A.  $\text{CrO}_4^{2-}$
- B.  $\text{SO}_3^{2-}$
- C.  $\text{H}^+$
- D.  $\text{SO}_4^{2-}$
- E.  $\text{Mn}^{2+}$

4. In a reaction between potassium chromate and sodium sulfite in acidified solution, which of the above will serve as the oxidizing agent?
5. Which of the above is added to the reactants side to denote that the reaction is occurring in acidic solution?
6. Forms a yellow color in solution

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**L Reisser**



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## **Table of Contents Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer**

1. Understanding the eBook Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
  - The Rise of Digital Reading Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
  - Advantages of eBooks Over Traditional Books
2. Identifying Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
  - User-Friendly Interface
4. Exploring eBook Recommendations from Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
  - Personalized Recommendations
  - Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer User Reviews and Ratings
  - Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer and Bestseller Lists
5. Accessing Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer Free and Paid eBooks
  - Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer Public Domain eBooks
  - Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer eBook Subscription Services
  - Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer Budget-Friendly Options

6. Navigating Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer eBook Formats
  - ePub, PDF, MOBI, and More
  - Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer Compatibility with Devices
  - Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
  - Highlighting and Note-Taking Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
  - Interactive Elements Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
8. Staying Engaged with Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
9. Balancing eBooks and Physical Books Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
  - Setting Reading Goals Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
  - Fact-Checking eBook Content of Practice Exam Ap Chemistry Midterm 1st 9 Weeks Answer
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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