

Fictitious symbols are used for the first 18 elements in the periodic table. Use the clues below to write the fictitious symbol in the appropriate spot on the periodic table provided. Symbols for real elements do not represent those elements. HINT: You do not have to complete each clue in order.

1	2	13	14	15	16	17	18
X		A	T	H P	G	Q	L
U	R	AI	DP	Z	E	K	W
J	M						V

- Clue #1 U and J are alkali metals. J has more energy levels than U.
- Clue #2 I has 4 valence electrons on the ~~3rd~~^{2nd} energy level.
- Clue #3 M is a metal in period 3 with 2 valence electrons
- Clue #4 X has one proton in its nucleus
- Clue #5 Q has 2 energy levels and is a halogen
- Clue #6 L is the only noble gas that doesn't have 8 valence electrons
- Clue #7 A is the only metalloid in the 13th group.
- Clue #8 I has 3 valence electrons on the third energy level
- Clue #9 V and W are un-reactive elements. V has more electrons than W.
- Clue #10 E and G have a common ion of -2 (E^{-2} , G^{-2}). E has 16 neutrons.
- Clue #11 R has an nuclear symbol of 3_1R
- Clue #12 K has one less electron than V
- Clue #13 D and Z are in the same period as J. However D is a metalloid and Z is a nonmetal.
- Clue #14 H has 7 neutrons

Periodic Puzzling Answer Key

Wolfgang Guggemos

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Periodic Puzzling Answer Key

1. Understanding the eBook Periodic Puzzling Answer Key
 - The Rise of Digital Reading Periodic Puzzling Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Puzzling Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Puzzling Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Puzzling Answer Key
 - Personalized Recommendations
 - Periodic Puzzling Answer Key User Reviews and Ratings
 - Periodic Puzzling Answer Key and Bestseller Lists
5. Accessing Periodic Puzzling Answer Key Free and Paid eBooks
 - Periodic Puzzling Answer Key Public Domain eBooks
 - Periodic Puzzling Answer Key eBook Subscription Services
 - Periodic Puzzling Answer Key Budget-Friendly Options

6. Navigating Periodic Puzzling Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Puzzling Answer Key Compatibility with Devices
 - Periodic Puzzling Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Puzzling Answer Key
 - Highlighting and Note-Taking Periodic Puzzling Answer Key
 - Interactive Elements Periodic Puzzling Answer Key
8. Staying Engaged with Periodic Puzzling Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Puzzling Answer Key
9. Balancing eBooks and Physical Books Periodic Puzzling Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Puzzling Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Puzzling Answer Key
 - Setting Reading Goals Periodic Puzzling Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Puzzling Answer Key
 - Fact-Checking eBook Content of Periodic Puzzling Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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