

% Composition & Mass %: Multiple Choice

1. The percentage of nitrogen, by mass, in $\text{Ca}(\text{CN})_2$ is:

- A) 14.9% B) 26.0% C) 30.4%
D) 43.6% E) None of the above

ANS: C

1. The percentage of calcium, by mass, in $\text{Ca}(\text{CN})_2$ is:

- A) 14.9% B) 26.0% C) 30.4%
D) 43.6% E) None of the above

ANS: D

1. The percentage of carbon, by mass, in $\text{Ca}(\text{CN})_2$ is:

- A) 14.9% B) 26% C) 30.4%
D) 43.6% E) None of the above

ANS: B

1. The percentage composition by mass of each element in $\text{Ca}(\text{CN})_2$ is:

- A) 32.4% Ca, 22.6% C, 45.0% N B) 22.6% Ca, 32.4% C, 45.0% N
C) 43.6% Ca, 26% C, 30.4% N D) 20.7% Ca, 43.7% C, 35.6% N
E) 35.0% Ca, 20.0% C, 45.0% N

ANS: C

1. The percentage of magnesium, by mass, in $\text{Mg}_3(\text{PO}_4)_2$ is:

- A) 15.9% B) 23.6% C) 27.7%
D) 48.7% E) None of the above

ANS: C

1. The percentage of phosphorus, by mass, in $\text{Mg}_3(\text{PO}_4)_2$ is:

- A) 15.9% B) 23.6% C) 27.7%
D) 48.7% E) None of the above

ANS: B

1. The percentage of oxygen, by mass, in $\text{Mg}_3(\text{PO}_4)_2$ is:

- A) 15.9% B) 23.6% C) 27.7%
D) 48.7% E) None of the above

ANS: D

Percent Composition Multiple Choice

Mr. Ajeet Singh



Percent Composition Multiple Choice:

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Cambridge Checkpoints VCE Chemistry Units 3 and 4 2013 Roger Slade, Maureen Slade, 2012-12-14

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National Research Council, Board on Testing and Assessment, Center for Education, Committee on NAEP Reporting Practices, 2001-02-21 At the request of the Department of Education the National Research Council formed the Committee on NAEP Reporting Practices to address questions about the desirability feasibility and potential impact of implementing these reporting practices The committee developed study questions designed to address issues surrounding district level and market basket reporting

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A New Approach to ICSE Chemistry for Class IX (A.Y. 2023-24) Onward V. K. Sally, 2023-05-20 The book A New Approach to I C S E Chemistry for Class IX has been written in accordance with the latest syllabus prescribed by the Council for Indian School Certificate Examination New Delhi The book is divided into five distinct parts 1 The first part constitutes chapters on The Languages of chemistry Chemical Changes and Reactions and Water to give an idea to the students about the diversity of matter 2 The second part constitutes chapters on Atomic Structure and Chemical Bonding and The Periodic Table This part will help the students to understand unity in diversity of matter It further explains the rules required for reaction between various forms of matter 3 The third part constitutes chapter on Study of the First Element Hydrogen and Study of Gas Laws This will familiarise the students with element hydrogen and their chemical properties It further explains the behaviour of gases under changes of temperature and pressure and also provides explanation in terms of molecular motion 4 The fourth part deals with Atmospheric Pollution 5 The fifth part deals with Practical Chemistry The salient features of the book Large number of chemical reactions are described with experimental observations Important points have been highlighted for location of precise answers All definitions and other important points are highlighted Diagrams are made simple and more Informative In the chapter on the language of chemistry a new method has been introduced for balancing chemical equations This method Is based on atomic numbers which Is fundamental property of the elements A chapter on Practical Chemistry is added to help students for laboratory work Questions have been added chapter wise under the heading Question Bank for the benefit of students The

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