

Use the partial periodic table to answer questions 4 -7.

⁶
C
Carbon
12.0107

— Atomic number
 — Symbol
 — Name
 — Average Atomic Mass

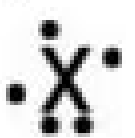
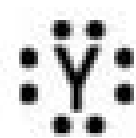
Partial Periodic Table of the Elements

1A	1	H	Hydrogen	1.00794
2A	2	He	Helium	4.0026
3A	3	Li	Lithium	6.941
4A	4	Be	Beryllium	9.0122
5A	5	B	Boron	10.811
6A	6	C	Carbon	12.0107
7A	7	N	Nitrogen	14.0067
8A	8	O	Oxygen	15.9994
9A	9	F	Fluorine	18.9984
10A	10	Ne	Neon	20.1797
11A	11	Na	Sodium	22.9897
12A	12	Mg	Magnesium	24.3050
13A	13	Al	Aluminum	26.98154
14A	14	Si	Silicon	28.0855
15A	15	P	Phosphorus	30.97376
16A	16	S	Sulfur	32.065
17A	17	Cl	Chlorine	35.453
18A	18	Ar	Argon	39.948

1A	2A	3A	4A	5A	6A	7A	8A
1	2	3	4	5	6	7	8
H	He	Li	Be	B	C	N	O
Hydrogen	Helium	Lithium	Beryllium	Boron	Carbon	Nitrogen	Oxygen
1.00794	4.0026	6.941	9.0122	10.811	12.0107	14.0067	15.9994
Na	Mg	Al	Si	P	S	Cl	Ar
Sodium	Magnesium	Aluminum	Silicon	Phosphorus	Sulfur	Chlorine	Argon
22.9897	24.3050	26.98154	28.0855	30.97376	32.065	35.453	39.948

Question 5

Suppose scientists discovered four new elements (W, X, Y, Z) while studying rock and soil samples brought back from a Mars mission. Which Lewis dot structure represents an element that should be placed in column VIIA (17) of the periodic table?

- A.  B.  C.  D. 

Partial Periodic Answer Physical Science Lab

G Thomas



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