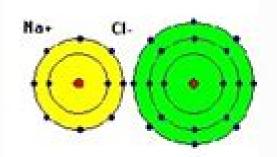
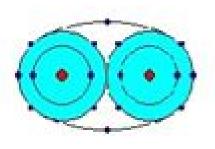
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- In <u>ionic bonding</u>, atoms <u>transfer</u> electrons to achieve noble gas configuration.
- In <u>covalent bonding</u>, atoms <u>share</u> electrons to achieve noble gas configuration.
- Most atoms <u>share</u> electrons until they have a total of <u>8</u> valence electrons (<u>octet</u> rule). However, <u>hydrogen</u> only needs <u>2</u> electrons to be stable.





Section 8 2 The Nature Of Covalent Bonding Workbook

DJ Losen

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