

### Balancing Chemical Equations Worksheet

1. 2  $\text{H}_2$  + 1  $\text{O}_2 \rightarrow$  2  $\text{H}_2\text{O}$
2. 1  $\text{N}_2$  + 3  $\text{H}_2 \rightarrow$  2  $\text{NH}_3$
3. 1  $\text{S}_8$  + 12  $\text{O}_2 \rightarrow$  8  $\text{SO}_3$
4. 2  $\text{N}_2$  + 1  $\text{O}_2 \rightarrow$  2  $\text{N}_2\text{O}$
5. 2  $\text{HgO} \rightarrow$  2  $\text{Hg}$  + 1  $\text{O}_2$
6. 6  $\text{CO}_2$  + 6  $\text{H}_2\text{O} \rightarrow$  1  $\text{C}_6\text{H}_{12}\text{O}_6$  + 6  $\text{O}_2$
7. 1  $\text{Zn}$  + 2  $\text{HCl} \rightarrow$  1  $\text{ZnCl}_2$  + 1  $\text{H}_2$
8. 1  $\text{SiCl}_4$  + 4  $\text{H}_2\text{O} \rightarrow$  1  $\text{H}_4\text{SiO}_4$  + 4  $\text{HCl}$
9. 2  $\text{Na}$  + 2  $\text{H}_2\text{O} \rightarrow$  2  $\text{NaOH}$  + 1  $\text{H}_2$
10. 2  $\text{H}_3\text{PO}_4 \rightarrow$  1  $\text{H}_4\text{P}_2\text{O}_7$  + 1  $\text{H}_2\text{O}$
11. 1  $\text{C}_{10}\text{H}_{16}$  + 8  $\text{Cl}_2 \rightarrow$  10  $\text{C}$  + 16  $\text{HCl}$
12. 1  $\text{CO}_2$  + 2  $\text{NH}_3 \rightarrow$  1  $\text{OC}(\text{NH}_2)_2$  + 1  $\text{H}_2\text{O}$
13. 4  $\text{Si}_2\text{H}_3$  + 17  $\text{O}_2 \rightarrow$  8  $\text{SiO}_2$  + 6  $\text{H}_2\text{O}_3$
14. 2  $\text{Al}(\text{OH})_3$  + 3  $\text{H}_2\text{SO}_4 \rightarrow$  1  $\text{Al}_2(\text{SO}_4)_3$  + 6  $\text{H}_2\text{O}$
15. 4  $\text{Fe}$  + 3  $\text{O}_2 \rightarrow$  2  $\text{Fe}_2\text{O}_3$
16. 1  $\text{Fe}_2(\text{SO}_4)_3$  + 6  $\text{KOH} \rightarrow$  3  $\text{K}_2\text{SO}_4$  + 2  $\text{Fe}(\text{OH})_3$
17. 2  $\text{C}_7\text{H}_6\text{O}_2$  + 15  $\text{O}_2 \rightarrow$  14  $\text{CO}_2$  + 6  $\text{H}_2\text{O}$
18. 1  $\text{H}_2\text{SO}_4$  + 8  $\text{HI} \rightarrow$  1  $\text{H}_2\text{S}$  + 4  $\text{I}_2$  + 4  $\text{H}_2\text{O}$
19. 4  $\text{FeS}_2$  + 11  $\text{O}_2 \rightarrow$  2  $\text{Fe}_2\text{O}_3$  + 8  $\text{SO}_2$
20. 2  $\text{Al}$  + 3  $\text{FeO} \rightarrow$  1  $\text{Al}_2\text{O}_3$  + 3  $\text{Fe}$
21. 1  $\text{Fe}_2\text{O}_3$  + 3  $\text{H}_2 \rightarrow$  2  $\text{Fe}$  + 3  $\text{H}_2\text{O}$
22. 1  $\text{Na}_2\text{CO}_3$  + 2  $\text{HCl} \rightarrow$  2  $\text{NaCl}$  + 1  $\text{H}_2\text{O}$  + 1  $\text{CO}_2$
23. 2  $\text{K}$  + 1  $\text{Br}_2 \rightarrow$  2  $\text{KBr}$
24. 1  $\text{C}_7\text{H}_{16}$  + 11  $\text{O}_2 \rightarrow$  7  $\text{CO}_2$  + 8  $\text{H}_2\text{O}$
25. 1  $\text{P}_4$  + 5  $\text{O}_2 \rightarrow$  2  $\text{P}_2\text{O}_5$

# Section Reactions And Equations Key

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## Section Reactions And Equations Key:

**Survival Guide to General Chemistry** Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step by step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid base reactions, which often persists through working with acid base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields. NBS Special Publication, 1968 General, Organic, and Biochemistry Media Update Ira Blei, George Odian, 2008-12-19 Blei and Odian's text gives students the tools they need to develop a working understanding of chemical principles rather than just asking them to memorize facts. Now available in a new media enhanced version complete with its own online course space learning environment ChemPortal. Blei Odian is better suited than ever to meet the needs of the students taking this course. The Media Update version of Blei Odian includes references to dynamic interactive tutorials which provide a step by step walkthrough of concepts and problem solving skills as well as answer specific feedback and practice problems. We recognize that all introductory courses are not alike. For that reason, we offer this text in three versions so you can choose the option that's right for you. General Organic and Biochemistry cloth 0 7167 4375 2 paper 1 4292 0994 1 the comprehensive 26 chapter text An Introduction to General Chemistry 0 7167 7073 3 10 chapters that cover the core concepts in general chemistry Organic and Biochemistry 0 7167 7072 5 16 chapters that cover organic and biochemistry plus two introductory chapters that review general chemistry **Chemistry: The Central Science** Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalus Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text, think again. In response to market request, we have created the third Australian edition of the US bestseller Chemistry: The Central Science. An extensive revision has taken this text to new heights. Triple checked for scientific accuracy and consistency, this edition is a more seamless and cohesive product yet retains the clarity, innovative pedagogy, functional problem solving, and visuals of the previous version. All artwork and images are now consistent in quality across the entire text. And with a more traditional and

logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation     *Publications of the National Institute of Standards and Technology ... Catalog* National Institute of Standards and Technology (U.S.),1981

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Efficiently Studying Organic Chemistry Eberhard Breitmaier,2022-04-11 Efficiently Studying Organic Chemistry Complete yet concise learning resource for organic chemistry exam training Based on the author s extensive teaching experience this unique textbook comprises the essentials of organic chemistry in 86 chapters as concise self contained units of study Each chapter visually presented as one or two double pages includes questions to allow for immediate and effective self examination Answers are summarized in the appendix Topics covered within the book include Basic concepts atomic and molecular orbitals covalent bonding hybridization resonance aromaticity Molecular structure atom connectivity skeletal isomerism conformation configuration chirality The classes of organic compounds including natural products polymers and biopolymers Types mechanisms selectivity and specificity of organic reactions Molecular structure elucidation mass spectrometry UV and visible light absorption IR and NMR spectroscopy Planning organic syntheses The perfect fit for bachelor and master students alike this book is an all in one resource for efficiently studying and passing organic chemistry exams

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