

### Module 3: Elements and the Periodic Table

#### Directions

Click on the following website: [www.ptable.com](http://www.ptable.com)

You will interact with this periodic table to find the answers to the questions below.

#### Part 1; Atomic Numbers and Atomic Masses

Complete the following table, by finding the atomic numbers and atomic masses for each element:

| Symbol        | P      | S       | Cl     | Ar     | K       | Ca     |
|---------------|--------|---------|--------|--------|---------|--------|
| Atomic number | 15     | 14      | 17     | 18     | 19      | 20     |
| Atomic Mass   | 30.974 | 28.0855 | 35.453 | 39.948 | 39.0983 | 40.078 |

- Using the data in the table above, is the periodic table organized by increasing atomic mass? Explain. Yes, because higher atomic numbers mean more subatomic particles which increase mass
- What property is the periodic table organized by? Atomic number

#### Part 2: Groups and Periods:

Continue to work in this site: [www.ptable.com](http://www.ptable.com)

Make sure the "Wikipedia" tab is selected in order to answer the following questions:

|                                                                                                                            | Write your answers here: |
|----------------------------------------------------------------------------------------------------------------------------|--------------------------|
| 1. The vertical columns on the periodic table are called <i>groups</i> , how many groups are there on the periodic table?  | 16                       |
| 2. The horizontal rows on the periodic table are called <i>periods</i> , how many periods are there on the periodic table? | 7                        |
| 3. Which element is located in group 10 and period 5                                                                       | Palladium                |
| 4. Which element is located in group 15 and period 4                                                                       | Arsenic                  |
| 5. Which element is located in group 2 and period 3                                                                        | Magnesium                |
| 6. Which element is located in group 18 and period 6                                                                       | Radon                    |
| 7. Which element is located in group 1 and period 7                                                                        | Francium                 |
| 8. Are most of the elements on the periodic table classified as a metal or a non-metal?                                    | Metals                   |
| 9. Are the non-metal elements located on the left or right side of the periodic table?                                     | Right                    |

#### Part 3: Families

## Section 3 The Periodic Table Answers

**Patrick Vollmar**



### Section 3 The Periodic Table Answers:

**Regents Exams and Answers: Chemistry--Physical Setting Revised Edition** Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day

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### **Table of Contents Section 3 The Periodic Table Answers**

1. Understanding the eBook Section 3 The Periodic Table Answers
  - The Rise of Digital Reading Section 3 The Periodic Table Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Section 3 The Periodic Table Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an eBook Section 3 The Periodic Table Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Section 3 The Periodic Table Answers
  - Personalized Recommendations
  - Section 3 The Periodic Table Answers User Reviews and Ratings
  - Section 3 The Periodic Table Answers and Bestseller Lists

5. Accessing Section 3 The Periodic Table Answers Free and Paid eBooks
  - Section 3 The Periodic Table Answers Public Domain eBooks
  - Section 3 The Periodic Table Answers eBook Subscription Services
  - Section 3 The Periodic Table Answers Budget-Friendly Options
6. Navigating Section 3 The Periodic Table Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Section 3 The Periodic Table Answers Compatibility with Devices
  - Section 3 The Periodic Table Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Section 3 The Periodic Table Answers
  - Highlighting and Note-Taking Section 3 The Periodic Table Answers
  - Interactive Elements Section 3 The Periodic Table Answers
8. Staying Engaged with Section 3 The Periodic Table Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Section 3 The Periodic Table Answers
9. Balancing eBooks and Physical Books Section 3 The Periodic Table Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Section 3 The Periodic Table Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Section 3 The Periodic Table Answers
  - Setting Reading Goals Section 3 The Periodic Table Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Section 3 The Periodic Table Answers
  - Fact-Checking eBook Content of Section 3 The Periodic Table Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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