

CHAPTER 11 REVIEW*Molecular Composition of Gases***MIXED REVIEW****SHORT ANSWER** Answer the following questions in the space provided.

1. _____ The average speed of a gas molecule is most directly related to the _____.
 - (a) polarity of the molecule
 - (b) pressure of the gas
 - (c) temperature of the gas
 - (d) number of moles in the sample
2. _____ Which of the following statements best represents Graham's law for two gases at the same temperature?
 - (a) The molar masses and the velocities are inversely proportional.
 - (b) The molar masses and the velocities are directly proportional.
 - (c) The molar masses and the square roots of the velocities are directly proportional.
 - (d) The molar masses and the squares of the velocities are inversely proportional.
3. Water is synthesized from its elements according to the following equation:
$$2\text{H}_2(g) + \text{O}_2(g) \rightarrow 2\text{H}_2\text{O}(l)$$
 - _____ a. Which is greater, the volume of H_2 consumed or the volume of O_2 consumed?
 - _____ b. Which is greater, the mass of H_2 consumed or the mass of O_2 consumed?
4. Consider a 2.24 L sample of an ideal gas at STP.
 - _____ a. How many moles are present in the sample, or is more data needed?
 - _____ b. How many molecules are present in the sample, or is more data needed?
 - _____ c. What is the mass of the sample, or is more data needed?

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

5. _____ Calculate the volume of 3.5 mol of an ideal gas at 24°C and 760 torr.

Section 3 Molecular Composition Of Gases Answers

Arshad Iqbal



Section 3 Molecular Composition Of Gases Answers:

Answer Book National Geographic Society (U.S.), 2009 Covers everything from earth sciences to astronomy from climate and habitats to human arts and cultures from ancient history to cutting edge technology and descriptions flags and statistics of all the countries in the world

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A Pronouncing Chemical Formula Speller and Contest Guide Carl Alfred Jacobson,1925

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1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Section 3 Molecular Composition Of Gases Answers
 - Chapter 2: Essential Elements of Section 3 Molecular Composition Of Gases Answers
 - Chapter 3: Section 3 Molecular Composition Of Gases Answers in Everyday Life
 - Chapter 4: Section 3 Molecular Composition Of Gases Answers in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of Section 3 Molecular Composition Of Gases Answers. The first chapter will explore what Section 3 Molecular Composition Of Gases Answers is, why Section 3 Molecular Composition Of Gases Answers is vital, and how to effectively learn about Section 3 Molecular Composition Of Gases Answers.
 3. In chapter 2, this book will delve into the foundational concepts of Section 3 Molecular Composition Of Gases Answers. The second chapter will elucidate the essential principles that need to be understood to grasp Section 3 Molecular Composition Of Gases Answers in its entirety.
 4. In chapter 3, the author will examine the practical applications of Section 3 Molecular Composition Of Gases Answers in daily life. This chapter will showcase real-world examples of how Section 3 Molecular Composition Of Gases Answers can be effectively utilized in everyday scenarios.
 5. In chapter 4, this book will scrutinize the relevance of Section 3 Molecular Composition Of Gases Answers in specific contexts. The fourth chapter will explore how Section 3 Molecular Composition Of Gases Answers is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, this book will draw a conclusion about Section 3 Molecular Composition Of Gases Answers. This chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Section 3 Molecular Composition Of Gases Answers.

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