

Chapter 7 Chemical Reactions

6. What is a double-replacement reaction? A double-replacement reaction is a reaction in which two different compounds exchange positive ions and form two new compounds.
7. Complete the chart by filling in the general forms of the reactions shown.

General Forms	
Single-Replacement Reaction	Double-Replacement Reaction
$A + BC \longrightarrow A + AC$	$AB + CD \longrightarrow AD + CB$

8. Lead(II) nitrate reacts with potassium iodide to form lead(II) iodide and potassium nitrate. Write the balanced equation for this double-replacement reaction. $Pb(NO_3)_2 + 2KI \longrightarrow PbI_2 + 2KNO_3$
9. Circle the letter of the correct answer. Calcium carbonate, $CaCO_3$, reacts with hydrochloric acid, HCl , in a double-replacement reaction. What are the products of this reaction?
- ☒ a. calcium chloride, $CaCl_2$, and carbonic acid, H_2CO_3
☐ b. calcium hydride, CaH_2 , chlorine, Cl_2 , and carbon dioxide, CO_2
☐ c. calcium hydrogen carbonate, $Ca(HCO_3)_2$, and chlorine, Cl_2
☐ d. calcium perchlorate, $Ca(ClO_4)_2$, and methane, CH_4
10. Is the following sentence true or false? A combustion reaction is a reaction in which a substance reacts with carbon dioxide, often producing heat and light. false
11. Methane, CH_4 , burns in oxygen to form carbon dioxide and water. Write the balanced equation for this reaction. $CH_4 + 2O_2 \longrightarrow CO_2 + 2H_2O$
12. Is the following sentence true or false? The reaction that forms water can be classified as either a synthesis reaction or a combustion reaction. true

Reactions as Electron Transfers (pages 204–205)

13. What is an oxidation-reduction reaction? An oxidation-reduction reaction is a reaction in which electrons are transferred from one reactant to another.
14. Calcium reacts with oxygen to form calcium oxide. Which reactant is oxidized in this reaction? calcium
15. Is the following sentence true or false? When calcium reacts with oxygen, each calcium atom gains two electrons and becomes a calcium ion with a charge of $2+$. false
16. Is the following sentence true or false? Oxygen must be present in order for an oxidation-reduction reaction to take place. false
17. The process in which an element gains electrons during a chemical reaction is called reduction.

Prentice Hall Chemistry Hydrocarbon Answers

JS Bruner



Prentice Hall Chemistry Hydrocarbon Answers:

Prentice Hall Chemistry, 2000 *Introductory Organic Chemistry and Hydrocarbons* Caio Lima Firme, 2019-08-28 A novel proposal for teaching organic chemistry based on a broader and simplified use of quantum chemistry theories and notions of some statistical thermodynamic concepts aiming to enrich the learning process of the organic molecular properties and organic reactions A detailed physical chemistry approach to teach organic chemistry for undergraduate students is the main aim of this book A secondary objective is to familiarize undergraduate students with computational chemistry since most of illustrations of optimized geometries plus some topological graphs and information is from quantum chemistry outputs which will also enable students to obtain a deeper understanding of organic chemistry **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction [JEE Main Chemistry - Unit wise Practice Test Papers](#) Career Point Kota, 2020-07-19 Competitive examination preparation takes enormous efforts time on the part of a student to learn practice and master each unit of the syllabus To check proficiency level in each unit student must take self assessment to identify his her weak areas to work upon that eventually builds confidence to win Also performance of a student in exam improves significantly if student is familiar with the exact nature type and difficulty level of the questions being asked in the Exam With this objective in mind we are presenting before you this book containing unit tests Some features of the books are The complete syllabus is divided into logical units and there is a self assessment tests for each unit Tests are prepared by subject experts who have decade of experience to prepare students for competitive exams Tests are as per the latest pattern of the examination Detailed explanatory solution of each test paper is also given Student is advised to attempt these Tests once they complete the preparation revision of unit They should attempt these Test in exam like environment in a specified time Student is advised to properly analyze the solutions and think of alternative methods and linkage to the solutions of identical problems also We firmly believe that the book in this form will definitely help a genuine hardworking student We have put our best efforts to make this book error free still there may be some errors We would appreciate if the same is brought to our notice We wish to utilize the opportunity to place on record our special thanks to all faculty members and editorial team for their efforts to make this book *Computational Chemistry* Errol G. Lewars, 2024-06-04 This is the fourth edition of the successful textbook on computational chemistry which continues to provide a comprehensive introduction to the theory and practice of computational chemistry Notable updates include a review of references up to mid 2023 encompassing recent developments in scientific journals books and software The evolving prominence of density functional theory DFT is

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