

11-1 Practice Problems

1. Lead will react with Hydrochloric acid to produce Lead (II) Chloride and Hydrogen gas. How many moles of Hydrochloric acid are needed to completely react with 0.36-mol of lead?

2. How many moles of HNO_3 will be produced when 0.51-mol of N_2O_5 reacts according to the following equation?
 $\text{N}_2\text{O}_5 + \text{H}_2\text{O} \rightarrow 2\text{HNO}_3$

3. Iron will react with Oxygen Gas to produce Fe_2O_3 . How many moles of Fe_2O_3 will be produced if 0.18-mol of Fe reacts?

4. How many moles of NaBr will be produced when 0.69-mol of Bromine reacts according to the following equation?
 $\text{Br}_2 + 2\text{NaI} \rightarrow 2\text{NaBr} + \text{I}_2$

5. Phosphorus will react with Bromine to produce Phosphorus Tribromide. How many moles of Phosphorus Tribromide will be produced if 0.78-mol of Bromine is reacted?

6. Carbon will react with Zinc Oxide to produce Zinc and Carbon Dioxide. How many moles of carbon dioxide will be produced if 0.38-mol of ZnO is completely reacted?

7. How many moles of oxygen will be needed to react with 0.38-mol of C_3H_8 according to the following equation?
 $\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$

8. Nitrogen gas can react with Hydrogen gas to produce ammonia. How many moles of Nitrogen will be needed to produce 0.48-mol of NH_3 ?

9. How many moles of hydrogen will be produced if 0.44-mol of CaH_2 reacts according to the following equation?
 $\text{CaH}_2 + 2\text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2 + 2\text{H}_2$

10. How many moles of water will be produced if 2.35-mol of oxygen reacts according to the following equation?
 $2\text{C}_6\text{H}_6 + 15\text{O}_2 \rightarrow 12\text{CO}_2 + 6\text{H}_2\text{O}$

Prentice Hall Chemistry 11 2 Practice Problems Answers

Yan Bai



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Prentice Hall Chemistry, 2000 Science Education Research and Practice in Europe Doris Jorde, Justin Dillon, 2012-12-17 Each volume in the 7 volume series *The World of Science Education* reviews research in a key region of the world These regions include North America South and Latin America Asia Australia and New Zealand Europe Arab States and Sub Saharan Africa The focus of this Handbook is on science education in Europe In producing this volume the editors have invited a range of authors to describe their research in the context of developments in the continent and further afield In reading this book you are invited to consider the historical social and political contexts that have driven developments in science education research over the years A unique feature of science education in Europe is the impact of the European Union on research and development over many years A growing number of multi national projects have contributed to the establishment of a community of researchers increasingly accepting of methodological diversity That is not to say that Europe is moving towards homogeneity as this volume clearly shows Catalog of Copyright Entries. Third Series Library of Congress. Copyright Office, 1956 Includes Part 1 Number 2 Books and Pamphlets Including Serials and Contributions to Periodicals July December **Water Encyclopedia, Oceanography; Meteorology; Physics and Chemistry; Water Law; and Water History, Art, and Culture** Jay H. Lehr, 2005-06 Volume 1 outlines water supply infrastructure The requirements for supplying water to a home a city or a factory can be very different Experts in these fields explain the nuances of the details involved in maintaining adequate quantity and quality for these different consumers Waste water management can be of even greater concern yet its management can follow similar paths when compared to sophisticated water supply treatment Both the physics and chemistry of these fields are fully covered Volume 2 deals with the big picture of regional water supplies how they become contaminated how they can be protected and how they can best serve the surrounding populations and industries Significant focus is placed upon the natural chemistry of available water supplies and its biological impacts Case studies from regions around the world offer an excellent picture of the world s water resources *Criminalistics: Forensic Science, Crime, and Terrorism* James E. Girard, 2013-11-13 Designed for students that are not biology chemistry or physics majors this fully revised and updated Third Edition of the best selling *Criminalistics Forensic Science Crime and Terrorism* provides a comprehensive introduction to forensic science the scientific principles that are the underpinnings of crime analysis and the practical application of these principles Essential topics such as fingerprint identification DNA ballistics detection of forgeries forensic toxicology computer forensics and the identification and analysis of illicit drugs are thoroughly explained in a reader friendly manner Unlike comparable texts the Third Edition includes coverage of important terrorism and homeland security issues including explosives cybercrime cyberterrorism and weapons of mass destruction The text is also the only book on the market with a detailed description of DNA and CODIS techniques used by professionals *Addison-Wesley Chemistry* Antony C. Wilbraham, Dennis D. Staley, Michael S. Matta, Edward L. Waterman, Prentice-Hall

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