

Density Worksheet

Physical Science

$D = m/V$

| Densities of Common Substances @ 20°C | | | | |
|---------------------------------------|------------------------------|--|-----------|------------------------------|
| Substance | Density (g/cm ³) | | Substance | Density (g/cm ³) |
| Oxygen | 0.00133 | | Aluminum | 2.70 |
| Hydrogen | 0.000084 | | Iron | 7.87 |
| Ethanol | 0.785 | | Copper | 8.96 |
| Benzene | 0.880 | | Silver | 10.5 |
| Water | 1.000 | | Lead | 11.34 |
| Magnesium | 1.74 | | Mercury | 13.6 |
| Salt (sodium chloride) | 2.16 | | Gold | 19.32 |

1. The ratio of an object's mass to its _____ is called the *density* of the object.
2. A kilogram of lead occupies a much smaller volume than a kilogram of water, because _____ has a much higher *density*.
3. For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
 - a. mass = 453 g; volume = 225 cm³
 - b. mass = 5.0 g; volume = 10.0 cm³
 - c. mass = 26.1 g; volume = 2.0 mL
4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
5. A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

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Sebastian Brünink



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