

Use your notebook to complete all of the following review questions.
Please note that your completed notebook, corrected quizzes and unit review sheets are the **BEST** study tools. These questions will only help to reinforce some of the exam topics.

1. Draw a line continuum to show the metric prefixes in order from largest (on the left) to smallest (on the right).



2. Complete the following metric conversions:

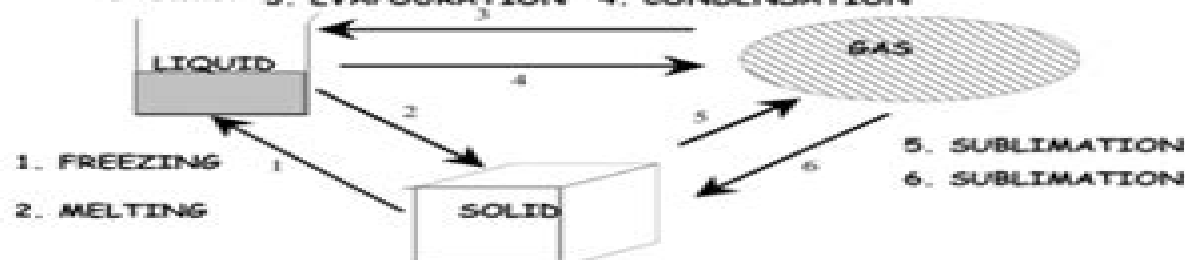
- a) 75 g = 0.075 kg e) 4.2 kL = 420 000 cL
 b) 4530 cm = 0.0453 km f) 83.2 cm = 832 mm
 c) 0.8 A = 800 mA g) 975 mg = 0.975 g
 d) 3200 mA = 3.2 A h) 975 mg = 0.000975 kg

3. Complete the following table using your knowledge of the Particle Theory of Matter:

STATE of MATTER	FORCES OF ATTRACTION	SPACING OF MOLECULES	MOVEMENT OF MOLECULES
solid	strongest	close together	least
liquid	weaker	farther apart	more
gas	weakest	farthest apart	most

4. Changes of state are examples of physical changes.

- a) Complete the following diagram with labels and arrows to show all of the SIX changes of state. 3. **EVAPORATION** 4. **CONDENSATION**



5. Name FIVE ways in which you know a chemical change has probably taken place.

colour change light given off
bubbles form heat produced
precipitate forms

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**University of Nebraska (Lincoln
campus). College of Engineering**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the text box.

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Table of Contents Physical Science Grade 9 Final Exam

1. Understanding the eBook Physical Science Grade 9 Final Exam
 - The Rise of Digital Reading Physical Science Grade 9 Final Exam
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Grade 9 Final Exam
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Grade 9 Final Exam
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Grade 9 Final Exam
 - Personalized Recommendations
 - Physical Science Grade 9 Final Exam User Reviews and Ratings
 - Physical Science Grade 9 Final Exam and Bestseller Lists

5. Accessing Physical Science Grade 9 Final Exam Free and Paid eBooks
 - Physical Science Grade 9 Final Exam Public Domain eBooks
 - Physical Science Grade 9 Final Exam eBook Subscription Services
 - Physical Science Grade 9 Final Exam Budget-Friendly Options
6. Navigating Physical Science Grade 9 Final Exam eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Grade 9 Final Exam Compatibility with Devices
 - Physical Science Grade 9 Final Exam Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Grade 9 Final Exam
 - Highlighting and Note-Taking Physical Science Grade 9 Final Exam
 - Interactive Elements Physical Science Grade 9 Final Exam
8. Staying Engaged with Physical Science Grade 9 Final Exam
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Grade 9 Final Exam
9. Balancing eBooks and Physical Books Physical Science Grade 9 Final Exam
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Grade 9 Final Exam
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Grade 9 Final Exam
 - Setting Reading Goals Physical Science Grade 9 Final Exam
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Grade 9 Final Exam
 - Fact-Checking eBook Content of Physical Science Grade 9 Final Exam
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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