ATOMIC STRUCTURE AND THE PERIODIC TABLE

Given the following isotopes, determine the atomic number, the mass number, the number electrons and neutrons.

| Symbol 1311 | Atomic Number | Mass Number | Protons | Electrons | Neutrons Isotope | | |
|----------------|------------------|----------------|---------|-----------|------------------|-----------------|--|
| 35S | 53 | | | | - cuirons | Isotope Name | |
| | 16 | 131 | 53 | 53 | 78 | iodine -13 | |
| 2He | | 35 | 16 | 16 | 19 | | |
| 27AI | 2 | 4 | 2 | 2 | | Julphur - 3 | |
| 81Kr | 13 | 27 | 13 | 12 | 2 | helium - 4 | |
| | 36 | 81 | | 15 | 14 | alvarinen - | |
| 81 37Rb | | | 36 | 36 | 45 | Krypten -8 | |
| | 37 | 13 | 37 | 37 | 44 | abdrum- | |

PARTA

Complete the following chart by writing the symbol for the isotope of the following elements. In addition, give the number of protons, electrons, mass number and atomic number and complete the

| Element | Neutrons | Protons | Electrons | Mass | Atomic | Isotope |
|------------|----------|---------|-----------|---------------|--------|-----------------|
| Uranium- | 145 | 92 | 92 | Number 237 | 92 | Symbol 23113 |
| Chlorine- | 28 | 17 | 17 | 45 | 17 | 4501 |
| Oxygen- | 9 | 7 | * | 17 | X | 1367 |
| Boron- | 6 | 5 | 5 | 11 | 5 | "R |
| Beryllium- | 5 | 4 | 4 | 9 | 11 | 1 PR. |
| Hydrogen- | 1 | 1 | 1 | 2 | 1 | 214 |
| Carbon- | 8 | 6 | 4 | 14 | 6 | 14C |

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Glen E. Moulton

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