## Section 7.4 - Determining Chemical Formulas

# Calculations of Molecular Formulas

An empirical formula may or may not be a correct molecular formula.

Book example , diborane's empirical formula is  $BH_3$ , any multiple of that equals the same ratio -  $B_2H_6$ ,  $B_3H_9$ ,  $B_4H_{12}$  etc

It is a colorless gas at room temperature with a repulsively sweet odor. Diborane mixes well with air, easily forming explosive mixtures. Diborane will ignite spontaneously in moist air at room temperature.

## **Section Quiz Determining Chemical Formulas**

**William Josephus Robinson** 

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