#### Respiration BMZ116B Study Guide

- To perform their many tasks, living cells require from outside sources.
- Energy enters most ecosystems as and leaves as .
- Photosynthesis generates \_\_\_\_ and \_\_\_\_ that the mitochondria of eukaryotes use as for cellular respiration.
- Cells harvest the chemical energy stored in organic molecules and use it to regenerate \_\_\_\_\_\_, the
  molecule that drives most cellular work.
- Respiration has three key pathways: \_\_\_\_\_, the \_\_\_\_\_, and \_\_\_\_\_.

#### A. The Principles of Energy Harvest

#### I. Cellular respiration and fermentation are catabolic, energy-yielding pathways.

- The arrangement of atoms of organic molecules represents energy.
- Enzymes \_\_\_\_\_ the systematic degradation of organic molecules that are rich in energy to simpler waste products with \_\_\_\_\_ energy.
- Some of the released energy is used to do \_\_\_\_\_; the rest is dissipated as \_\_\_\_\_.
- Catabolic metabolic pathways the energy stored in complex organic molecules.
- One type of catabolic process, \_\_\_\_\_\_, leads to the partial degradation of sugars in the absence of oxygen.
- A more efficient and widespread catabolic process. , consumes as a reactant to complete the breakdown of a variety of organic molecules.
  - In eukaryotic cells, are the site of most of the processes of cellular respiration.
- Cellular respiration is similar in broad principle to the combustion of gasoline in an automobile engine after \_\_\_\_\_\_ is mixed with hydrocarbon fuel.
  - is the fuel for respiration. The exhaust is \_\_\_\_\_ and \_\_\_\_.
- The overall process is:
  - ° organic compounds + \_\_\_\_ + \_\_\_ + energy (\_\_\_\_ +
- Carbohydrates, fats, and proteins can all be used as the fuel, but it is most useful to consider glucose.
   + 6O<sub>2</sub> → 6CO<sub>2</sub> + 6H<sub>2</sub>O + Energy (ATP + heat)
- The catabolism of glucose is \_\_\_\_\_ with a Δ G of -686 kcal per mole of glucose.
  - Some of this energy is used to produce , which can perform cellular work.

#### 2. Redox reactions release energy when electrons move closer to electronegative atoms.

- Catabolic pathways transfer the \_\_\_\_\_\_ stored in food molecules, releasing energy that is used to synthesize
- Reactions that result in the transfer of one or more electrons from one reactant to another are oxidation-reduction reactions, or \_\_\_\_\_\_\_\_\_reactions.
  - O The loss of electrons is called \_\_\_\_\_.
  - The addition of electrons is called \_\_\_\_\_\_\_.
- The formation of table salt from sodium and chloride is a redox reaction.

# **Section 9 Cellular Respiration Section Review Answer Key**

**Sabine Zange** 

#### **Section 9 Cellular Respiration Section Review Answer Key:**

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