

Name: _____

The Organization of the Periodic Table – Answer Key

Directions

Using your computer, tablet or mobile device, navigate to the website: www.ptable.com
You will interact with this periodic table to find the answers to the questions below.

*Make sure the "Wikipedia" tab is selected in order to answer the following questions:

1. The vertical columns on the periodic table are called *groups*, how many groups are there on the periodic table? **18**
2. The horizontal rows on the periodic table are called *periods*, how many periods are there on the periodic table? **7**
3. Using the group and period numbers, identify the elements that are located in each of the following location.
 - a. The element in group 10 and period 5 = **Pd**
 - b. The element in group 15 and period 4 = **As**
 - c. The element in group 2 and period 3 = **Mg**
 - d. The element in group 18 and period 6 = **Rn**
 - e. The element in group 1 and period 7 = **Fr**
4. Are most of the elements on the periodic table classified as a metal or a non-metal?
Metal
5. Are the non-metal elements located on the left or right side of the periodic table?
Right (except for Hydrogen)

*Elements can also be classified by *family*, use the color coding at the top of the page to help identify the family names and their element members.

6. How many elements belong to the Alkali Metal family? List the member elements by their symbol.
6 elements = Li, Na, K, Rb, Cs, Fr
7. Click on the group number for the Alkali Metals family at the top of the table. Answer the following about the Alkali metals:
 - a. What are 3 similar properties of these elements? **Shiny, soft, highly reactive**
 - b. Where are these found naturally? **Only in salts, not as a free element**
 - c. What is a common substance that alkali metals react vigorously with? **Water**

Section 3 The Periodic Table Answer Key

Léo Pomar

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Section 3 The Periodic Table Answer Key:

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Table of Contents Section 3 The Periodic Table Answer Key

1. Understanding the eBook Section 3 The Periodic Table Answer Key
 - The Rise of Digital Reading Section 3 The Periodic Table Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 3 The Periodic Table Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 3 The Periodic Table Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 3 The Periodic Table Answer Key

- Personalized Recommendations
- Section 3 The Periodic Table Answer Key User Reviews and Ratings
- Section 3 The Periodic Table Answer Key and Bestseller Lists
- 5. Accessing Section 3 The Periodic Table Answer Key Free and Paid eBooks
 - Section 3 The Periodic Table Answer Key Public Domain eBooks
 - Section 3 The Periodic Table Answer Key eBook Subscription Services
 - Section 3 The Periodic Table Answer Key Budget-Friendly Options
- 6. Navigating Section 3 The Periodic Table Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Section 3 The Periodic Table Answer Key Compatibility with Devices
 - Section 3 The Periodic Table Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 3 The Periodic Table Answer Key
 - Highlighting and Note-Taking Section 3 The Periodic Table Answer Key
 - Interactive Elements Section 3 The Periodic Table Answer Key
- 8. Staying Engaged with Section 3 The Periodic Table Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 3 The Periodic Table Answer Key
- 9. Balancing eBooks and Physical Books Section 3 The Periodic Table Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 3 The Periodic Table Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section 3 The Periodic Table Answer Key
 - Setting Reading Goals Section 3 The Periodic Table Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section 3 The Periodic Table Answer Key

- Fact-Checking eBook Content of Section 3 The Periodic Table Answer Key
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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