Name _____ Class _____ Date _____

9.1 THE ARITHMETIC OF EQUATIONS

SECTION REVIEW

Objectives

- Calculate the amount of reactants required or product formed in a nonchemical process.
- Interpret balanced chemical equations in terms of interacting moles, representative particles, masses, and gas volumes at STP.

Key Terms

- stoichiometry

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank may be completed with a term, short phrase, or number.

The coefficients of a balanced chemical equation indicate the relative number of ___1___ of reactants and products. All stoichiometric calculations begin with a ___2___. Only ___3___ and ___4___ are conserved in every reaction; moles, volumes, and representative particles may not be.

In solving stoichiometric problems, conversion factors relating moles of reactants or ___5___ of products are used.

If you assume ___6___, the equation also tells you about the volumes of gases.

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____

Part B True–False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

_____ 7. The coefficients in a balanced chemical equation can be used to derive mole ratios relating reactants to products.

_____ 8. The coefficients in a balanced chemical equation tell the relative volumes of reactants and products, expressed in any suitable units of volume.

_____ 9. To calculate the mass of a molecule in grams, you can use the molar mass and Avogadro's number.

Section 9.1 is on the reverse

Section 9 The Arithmetic Of Equations Answer Key

ML Yell

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web nov 4 2014 the invention relates to the field of grease production and more specifically relates to a formula of liquid shoe polish the liquid shoe polish comprises the following components in parts

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web the shoe polish was formulated in accordance to the procedure earlier on used in the production of shoe polish from carbon black gumel 2006 15g paraffin wax was melted at 970c to give a fine solution the temperature was lowered to 600 c 2g polyethylene pigment was added and stirred for 5 minutes

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web sep 4 2023 1 set up a double boiler fill a saucepan with 1 to 2 inches 2 54 to 5 08 centimeters of water place a heat safe bowl on top bring the water to a simmer over medium heat 2 add the olive oil and white beeswax into the bowl you will need about 2 8 ounces 79 38 grams of olive oil and 1 1 ounce 31 18 grams of white beeswax 1

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web apr 1 1996 a water resistant shoe polish including in addition to a tradition base mixture of wax and turpentine oil petroleum benzin high vacuum silicone grease glycerin glycerol stearate and stearic
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web the embodiment of the invention provides a method for producing shoe polish by using wastewater generated in the production of gallic acid which comprises the following steps adding calcium hydroxide suspension into wastewater from gallic acid production mixing standing and filtering to obtain filtrate

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web mar 6 2005 the functional group of the prepared shoe polish was also identified using ftir analysis and the ftir peaks confirmed the presence of a long chain ester group in the shoe polish

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