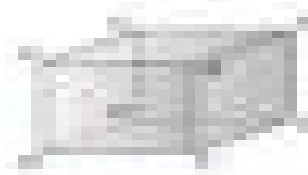


Answer following



- 1. Substrate and temperature: 20°C $100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$
- 2. A graph of initial velocity: $100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ $100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$
- 3. A graph of initial velocity: $100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ $100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$

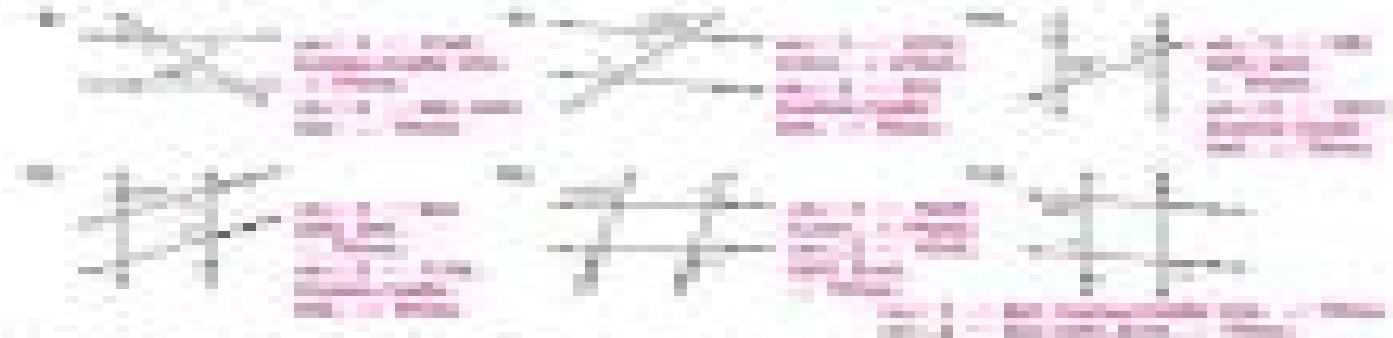
Calculate all values of K_m and V_{max} for the following reactions:
 (Note: All values are given in $\text{mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$)

- 1. $K_m = 100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ $V_{max} = 100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$
- 2. $K_m = 100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ $V_{max} = 100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$
- 3. $K_m = 100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ $V_{max} = 100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$
- 4. $K_m = 100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ $V_{max} = 100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$



Exercise 10

Graph the following data, and determine the K_m and V_{max} for the reaction.



Graph the following data, and determine the K_m and V_{max} for the reaction.



Prentice Hall Geometry 3 7 Enrichment Answers

Terry C. Jones



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