

Chapter 7 – Chemical Quantities and Reactions

7.1 THE MOLE

Collection Term // States a specific number of items.

- 1 dozen donuts = 12 donuts
- 1 ream of paper = 500 sheets
- 1 case = 24 cans

Mole // A collection that contains the same number of particles as there are carbon atoms in 12.0 g of carbon: 6.02×10^{23} atoms of an element.

Avogadro's Number // 6.02×10^{23} atoms of an element.

- 1 mole = 6.02×10^{23} atoms/molecules & this is an equality
- Used to convert moles of a substance to particles
- SIG FIGS count

Subscripts and Moles

- The subscripts in a formula show...
 - The relationship of atoms in the formula
 - The moles of each element in 1 mole of compound
- The subscripts are used to write conversion factors for moles of each element in 1 mole of a compound
 - 9 moles C/1 mole $C_9H_8O_4$
 - 8 moles H/1 mole $C_9H_8O_4$
 - 4 moles O/1 mole $C_9H_8O_4$

7.2-3 MOLAR MASS & CALCULATIONS

Molar Mass // The mass of 1 mole of an element or compound (atomic mass expressed in g).

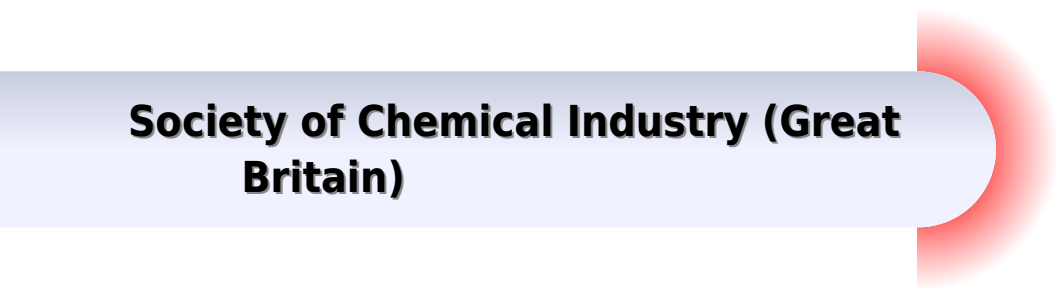
- For a compound...
 - The sum of the molar masses of the elements in the formula
 - $CaCl_2 = 40.1g/mol (Ca) + (2 \times 35.5g/mol) (Cl)$

Molar Mass Conversion Factors

- Are fractions written from the molar mass

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**Society of Chemical Industry (Great
Britain)**



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Computer Chemistry Mario Marsili, 2017-07-28 Computer Chemistry illustrates the methods and philosophies of how a computer can be instructed to understand chemical facts formulas and rules It focuses on discussions of all of the major sections in both theoretical framework and practical application through examples It includes the Synthesis Design Systems for the simulation of chemical reactions the Structure Elucidation Systems for the interpretation of spectral data the Molecular Modelling Systems for the visualization of chemical structures and the calculation of physico chemical parameters

The Pharmaceutical Era ,1893 *Chemical Matter* Prentice-Hall Staff,1994 Authorized teaching resource in Alberta for senior high science 14 24 1995 2004 Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Prentice Hall Chemistry** ,2000 *The Examiner* ,1840 **British Medical Journal** ,1919 The Saturday Review of Politics, Literature, Science and Art ,1887 **The Saturday Review of Politics, Literature, Science, Art, and Finance** ,1887 Nature Sir Norman Lockyer,1905 The Builder ,1899 **"The" Athenaeum** ,1835 *Seawater: Its Composition, Properties and Behaviour* John M. Wright,Angela Colling,2013-10-22 *Seawater Its Composition Properties and Behaviour* provides a comprehensive introduction to marine science This book is divided into seven chapters Chapter 1 summarizes the special properties of water and the role of the oceans in the hydrological cycle The distribution of temperature and salinity in the oceans and their combined influence on density stability and vertical water movements are discussed in Chapters 2 to 4 The fifth chapter describes the behavior of light and sound in seawater and provides examples of the application of acoustics to oceanography Chapter 6 examines the composition and behavior of the dissolved constituents of seawater covering minor and trace constituents and major ions as well as dissolved gases and biologically important nutrients Residence times speciation and carbonate equilibria are also deliberated The last chapter provides a short review of ideas about the history of seawater involvement of the oceans in global cycles and their relationship to climatic change This publication is beneficial to oceanographers and marine biologists including students that are interested in marine science **Athenaeum and Literary Chronicle** James Silk Buckingham,John Sterling,Frederick Denison Maurice,Henry Stebbing,Charles Wentworth Dilke,Thomas Kibble Hervey,William Hepworth Dixon,Norman Maccoll,Vernon Horace Rendall,John Middleton Murry,1835 **The Chemical News and Journal of Physical Science** ,1869 **tyhe educational times** ,1878 Athenaeum ,1845 **Seawater** Open Open University,1995-04-28 *Seawater* has been substantially updated in this second edition to take account of recent developments in marine science Sections dealing with difficult physical and chemical concepts have been

developed on the basis of feedback from the first edition making this an ideal learning tool for oceanography students Chapter 1 summarizes the special properties of water and the role of the oceans in the hydraulic cycle The distribution of temperature and salinity in the oceans and how they influence water density and movements is then discussed Light and sound in seawater are considered next along with some uses of acoustics These are followed by an examination of the composition and behaviour of dissolved constituents including such topics as residence times the control of pH and redox relationships Finally the history of seawater and its role in global cycles is reviewed with special reference to climatic change and the CO₂ problem Bulletin of the Atomic Scientists ,1966-06 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin s iconic Doomsday Clock stimulates solutions for a safer world **New England Journal of Education** ,1878

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