

Chemistry 12 Worksheet 2-2 LeChatelier's Principle

Unit 2 - Chemical Equilibrium

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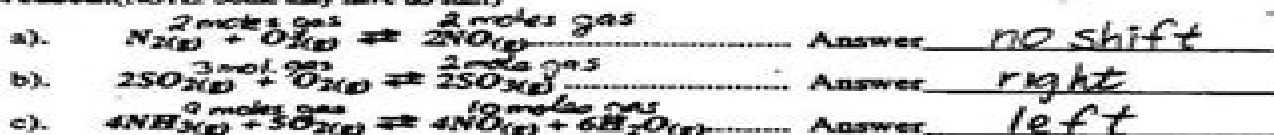
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TOTAL = 70 marks.

1. In order to decide what effect a change in total pressure will have on an equilibrium system with gases, what is the first thing you should do when given the balanced equation?

- Add up the moles of gas on each side of the equation.

2. Predict which way the following equilibrium systems will shift when the total pressure is increased. (NOTE: Some may have no shift)



3. Which way will the following equilibrium shift if the total pressure on the system is decreased?



It will shift right (more moles) to counteract the decrease in pressure.

Answer right

4. Explain why a flask filled with $\text{NO}_2(\text{g})$ and $\text{N}_2\text{O}_4(\text{g})$ will get darker when heated. Use the equation:



- heating will cause the equil^m to shift right.
This will produce more NO_2 which is dark brown. Hence the colour will get darker.

5. State Le Chatelier's Principle.

When a stress is applied to a system at equilibrium, the equil^m will shift (left or right) so as to counteract the imposed stress.

Prentice Hall Chemistry Chapter 7 Assessment Answers

**Klaus Gundertofte, Fleming Steen
Jørgensen**



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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Prentice Hall Chemistry**, 2000 **Molecular Modeling and Prediction of Bioactivity** Klaus Gundertofte, Fleming Steen Jørgensen, 2012-12-06 Much of chemistry molecular biology and drug design are centered around the relationships between chemical structure and measured properties of compounds and polymers such as viscosity acidity solubility toxicity enzyme binding and membrane penetration For any set of compounds these relationships are by necessity complicated particularly when the properties are of biological nature To investigate and utilize such complicated relationships henceforth abbreviated SAR for structure activity relationships and QSAR for quantitative SAR we need a description of the variation in chemical structure of relevant compounds and biological targets good measures of the biological properties and of course an ability to synthesize compounds of interest In addition we need reasonable ways to construct and express the relationships i e mathematical or other models as well as ways to select the compounds to be investigated so that the resulting QSAR indeed is informative and useful for the stated purposes In the present context these purposes typically are the conceptual understanding of the SAR and the ability to propose new compounds with improved property profiles Here we discuss the two latter parts of the SAR/QSAR problem i e reasonable ways to model the relationships and how to select compounds to make the models as good as possible The second is often called the problem of statistical experimental design which in the present context we call statistical molecular design SMD 1

Water-Insoluble Drug Formulation Ron Liu, 2018-03-12 Properties and Formulation From Theory to Real World Application Scientists have attributed more than 40 percent of the failures in new drug development to poor biopharmaceutical properties particularly water insolubility Issues surrounding water insolubility can postpone or completely derail important new drug development Even the much needed reformulation of currently marketed products can be significantly affected by these challenges More recently it was reported that the percentage increased to 90% for the candidates of new chemical entities in the discovery stage and 75% for compounds under development In the most comprehensive resource on the topic this third edition of Water Insoluble Drug Formulation brings together a distinguished team of experts to provide the scientific background and step by step guidance needed to deal with solubility issues in drug development Twenty three chapters systematically describe the detailed discussion on solubility theories solubility prediction models the aspects of preformulation biopharmaceutics pharmacokinetics regulatory and discovery support of water insoluble drugs to various techniques used in developing delivery systems for water insoluble drugs This book includes more

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