

7-3 Review and Reinforcement

Naming Chemical Compounds

Write the correct numerical prefix for each of the numbers in the table below.

Number	Prefix
1	mono
2	di
3	tri
4	tetra
5	penta
6	hexa
7	hepta
8	octa
9	nona
10	deca

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

cation
anion

hydrates
anhydrates

electronegative
hydrogen ions

ionic
molecular

11. Ionic compounds that absorb water into their solid structures form substances called **—hydrate—**.
12. When naming an ionic substance, the name of the **—cation—** comes first.
13. To name a binary molecular compound, add the suffix *-ide* to the more **—electronegative—** of the two elements.
14. By one definition, an acid is a molecular compound that dissolves in water to produce **—hydrogen ions—**.
15. The name of an acid is usually derived from the name of its characteristic **—anion—**.
16. Prefixes are often used when naming **molecular** compounds.

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