

## Preliminary Chemistry Notes - 2020

### Glossary of key terms:

**Decomposition** - The process by which organic substances are broken down into simpler organic matter

**Lustre** – How shiny a substance appears and relates to how well the substance is able to reflect light; in general, metals are lustrous, and non-metals are dull

**Matter** – The material substance that makes up the physical universe

**Molecule** - A group of two or more atoms covalently bonded together (usually refers to non-metals)

- Molecules can also be defined as particles which move independently of each other
- Thus, noble gases (group 8) which are chemically stable usually exists as single particles moving independently aka monoatomic molecules (containing one atom)
- Diatomic molecules are molecules with two atoms, e.g.  $H_2$ ,  $Cl_2$

**Compound** - Molecules that have at least two elements different, chemically combined in a specific ratio. They are represented by a chemical formula

**Mixture** – A substance of variable composition, containing two or more elements or compounds that are not chemically bonded together

**Covalent Bond** – A covalent bond, also called a molecular bond, is a chemical bond that involves the sharing of electron pairs between atoms

**Period** - A period in the periodic table is a row of chemical elements. All elements in a row have the same number of electron shells. Each next element in a period has one more proton and is less metallic than its predecessor

**Group** - The periodic table also has a special name for its vertical columns. Each column is called a group. The elements in each group have the same number of electrons in the outer orbital. Those outer electrons are also called valence electrons

**Ion** – Atom or molecule with a net electrical charge

**Polyatomic Ion** - A polyatomic ion, also known as a molecular ion, is a **charged** chemical species composed of two or more atoms covalently bonded

**Solution** – A homogeneous mixture of two or more substances, with a minor substance being uniformly distributed in a major substance. The minor substance is called the solute and the major substance is called the solvent. An example is saltwater (where salt = solute, and water = solvent). Also, the solute is dissolved in the solvent, and solute particles cannot be seen under a light microscope

# Preliminary Chemistry Notes

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## **Preliminary Chemistry Notes:**

**Excel Preliminary Chemistry** C. M. Roebuck, 2003      *Preliminary Chemistry Notes* ,2017-09      *Creative Chemistry Teaching Materials* Peter Barnes, 1995      **National Bureau of Standards Handbook** ,1964      *Educational Times* ,1916

**The Publishers' Circular and Booksellers' Record** ,1907      Education Outlook ,1914      **The Educational Times, and Journal of the College of Preceptors** ,1915      *The Academy and Literature* ,1902      **Journal of Education and School World** ,1889      B.H. Blackwell B.H. Blackwell Ltd, 1928      Publishers' Circular and Booksellers' Record of British and Foreign Literature ,1890      **The Journal of Education** ,1902      **Subject Index of Modern Books Acquired** British Library, 1918      **The Spectator** ,1902 A weekly review of politics literature theology and art      **The Publisher** ,1903

Catalogue of Scientific Papers Royal Society (Great Britain), 1914      **Catalogue of Scientific Papers (1800-1900): ser. 4 , 1884-1900** Royal Society (Great Britain), 1914      **The National Union Catalog, Pre-1956 Imprints** Library of Congress, American Library Association. Committee on Resources of American Libraries. National Union Catalog Subcommittee, 1969      **Bookseller and the Stationery Trades' Journal** ,1891

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## **Table of Contents Preliminary Chemistry Notes**

1. Understanding the eBook Preliminary Chemistry Notes
  - The Rise of Digital Reading Preliminary Chemistry Notes
  - Advantages of eBooks Over Traditional Books
2. Identifying Preliminary Chemistry Notes
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Preliminary Chemistry Notes
  - User-Friendly Interface
4. Exploring eBook Recommendations from Preliminary Chemistry Notes

- Personalized Recommendations
- Preliminary Chemistry Notes User Reviews and Ratings
- Preliminary Chemistry Notes and Bestseller Lists
- 5. Accessing Preliminary Chemistry Notes Free and Paid eBooks
  - Preliminary Chemistry Notes Public Domain eBooks
  - Preliminary Chemistry Notes eBook Subscription Services
  - Preliminary Chemistry Notes Budget-Friendly Options
- 6. Navigating Preliminary Chemistry Notes eBook Formats
  - ePub, PDF, MOBI, and More
  - Preliminary Chemistry Notes Compatibility with Devices
  - Preliminary Chemistry Notes Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Preliminary Chemistry Notes
  - Highlighting and Note-Taking Preliminary Chemistry Notes
  - Interactive Elements Preliminary Chemistry Notes
- 8. Staying Engaged with Preliminary Chemistry Notes
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Preliminary Chemistry Notes
- 9. Balancing eBooks and Physical Books Preliminary Chemistry Notes
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Preliminary Chemistry Notes
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Preliminary Chemistry Notes
  - Setting Reading Goals Preliminary Chemistry Notes
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Preliminary Chemistry Notes

- Fact-Checking eBook Content of Preliminary Chemistry Notes
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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