

# Periodic Trends Practice Worksheet

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Use the periodic table and your knowledge of the periodic trends to answer the following questions.

- ① Identify each element as a metal, metalloid, or nonmetal
  - Ⓐ germanium \_\_\_\_\_
  - Ⓑ zinc \_\_\_\_\_
  - Ⓒ phosphorous \_\_\_\_\_
  - Ⓓ lithium \_\_\_\_\_
- ② Which of the two species is larger in each of the following pairs?
  - Ⓐ  $\text{N}^{3-}$  or  $\text{F}^-$  \_\_\_\_\_
  - Ⓑ  $\text{Mg}^{2+}$  or  $\text{Ca}^{2+}$  \_\_\_\_\_
  - Ⓒ  $\text{Fe}^{2+}$  or  $\text{Fe}^{3+}$  \_\_\_\_\_
- ③ Which of the two species is smaller in each of the following pairs?
  - Ⓐ  $\text{K}^+$  or  $\text{Li}^+$  \_\_\_\_\_
  - Ⓑ  $\text{Au}^+$  or  $\text{Au}^{3+}$  \_\_\_\_\_
  - Ⓒ  $\text{P}^{3-}$  or  $\text{N}^{3-}$  \_\_\_\_\_
- ④ What atom has the LARGEST atomic radius?  
\_\_\_\_\_
- ⑤ What atom has the SMALLEST atomic radius?  
\_\_\_\_\_
- ⑥ Which is larger?
  - Ⓐ  $\text{Mg}$  or  $\text{Mg}^{2+}$  \_\_\_\_\_
  - Ⓑ  $\text{I}$  or  $\text{I}^-$  \_\_\_\_\_
  - Ⓒ  $\text{Cl}^-$  or  $\text{Br}^-$  \_\_\_\_\_
  - Ⓓ  $\text{F}^-$  or  $\text{Ne}$  \_\_\_\_\_
  - Ⓔ  $\text{N}^{3-}$  or  $\text{P}^{3-}$  \_\_\_\_\_
- ⑦ Put the following in order of increasing radius:  $\text{Br}^-$   $\text{Se}^{2-}$   $\text{Sr}^{2+}$   $\text{Kr}$   $\text{Rb}^+$
- ⑧ Circle the correct answer.
  - Ⓐ Bromine has a higher / lower ionization energy than Lead
  - Ⓑ Barium has a higher / lower ionization energy than Radon
  - Ⓒ Fluorine has a higher / lower ionization energy than Arsenic
  - Ⓓ Lithium has a higher / lower ionization energy than Cesium
- ⑨ Which of the two species is larger in each of the following pairs?
  - Ⓐ  $\text{Mg}$  or  $\text{Sr}$
  - Ⓑ  $\text{Be}$  or  $\text{N}$
  - Ⓒ  $\text{P}$  or  $\text{Cl}$
  - Ⓓ  $\text{Ni}$  or  $\text{Br}$
- ⑩ Arrange the following in order of increasing electronegativity:  $\text{S}$ ,  $\text{O}$ ,  $\text{Ne}$ ,  $\text{Al}$

# Periodic Trends Practice Problems With Answers

**A Gutmann**



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## **Table of Contents Periodic Trends Practice Problems With Answers**

1. Understanding the eBook Periodic Trends Practice Problems With Answers
  - The Rise of Digital Reading Periodic Trends Practice Problems With Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Practice Problems With Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Trends Practice Problems With Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Practice Problems With Answers
  - Personalized Recommendations
  - Periodic Trends Practice Problems With Answers User Reviews and Ratings
  - Periodic Trends Practice Problems With Answers and Bestseller Lists
5. Accessing Periodic Trends Practice Problems With Answers Free and Paid eBooks
  - Periodic Trends Practice Problems With Answers Public Domain eBooks
  - Periodic Trends Practice Problems With Answers eBook Subscription Services
  - Periodic Trends Practice Problems With Answers Budget-Friendly Options
6. Navigating Periodic Trends Practice Problems With Answers eBook Formats



- ePub, PDF, MOBI, and More
  - Periodic Trends Practice Problems With Answers Compatibility with Devices
  - Periodic Trends Practice Problems With Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Periodic Trends Practice Problems With Answers
    - Highlighting and Note-Taking Periodic Trends Practice Problems With Answers
    - Interactive Elements Periodic Trends Practice Problems With Answers
  8. Staying Engaged with Periodic Trends Practice Problems With Answers
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Periodic Trends Practice Problems With Answers
  9. Balancing eBooks and Physical Books Periodic Trends Practice Problems With Answers
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Periodic Trends Practice Problems With Answers
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Periodic Trends Practice Problems With Answers
    - Setting Reading Goals Periodic Trends Practice Problems With Answers
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Periodic Trends Practice Problems With Answers
    - Fact-Checking eBook Content of Periodic Trends Practice Problems With Answers
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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