

Finding the Percent Composition

- **Percent Composition** - The percentage (by mass) of each element in a compound.

Example 1:

Find the percent composition of red candies in a bag of 200 candies, if there are 26 red, and 174 blue candies.

- Total Candies: 200
- Red Candies: 26
- Blue Candies: 174

$$\frac{26 \text{ red}}{200 \text{ total}} \times 100 = 13.0 \% \text{ red}$$

Formula:

$$\frac{\text{Part}}{\text{Total}} \times 100$$

$$\frac{174 \text{ blue}}{200 \text{ total}} \times 100 = 87.0 \% \text{ blue}$$

Example 2:

Find the percent composition of MgCl_2 .

- Total mass MgCl_2 : 95.21 g
- Mass of Mg: 24.31 g
- Mass of Cl_2 : 70.91 g

$$\frac{24.31 \text{ g}}{95.21 \text{ g}} \times 100 = 25.53 \% \text{ Mg}$$

$$\frac{70.91 \text{ g}}{95.21 \text{ g}} \times 100 = 74.47 \% \text{ Cl}_2$$

Example 3:

Find the percent composition of H_2SO_4 .

- Total mass H_2SO_4 : 98.08 g
- Mass of H_2 : 2.02 g
- Mass of S: 32.07 g
- Mass of O_4 : 64.00 g

$$\frac{2.02 \text{ g}}{98.08 \text{ g}} \times 100 = 2.06 \% \text{ H}_2$$

$$\frac{32.07 \text{ g}}{98.08 \text{ g}} \times 100 = 32.70 \% \text{ S}$$

$$\frac{64.00 \text{ g}}{98.08 \text{ g}} \times 100 = 65.25 \% \text{ O}_4$$

Percentage Composition Answers With Work

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Percentage Composition Answers With Work:

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises **Access to Chemistry** Avril Higton, Mike Clemmet, Elaine Golding, Alan V Jones, 2007-10-31 This text has been specifically designed to prepare people with previously limited chemical knowledge for entrance into science related courses such as Foundation and Access courses which involve chemistry in higher education Until now there have been no texts available for use on these courses and this book fills that gap Access to Chemistry effectively forms a self study course which is split into separate modules and units covering the full spread of concepts required for those needing a basic knowledge of chemistry The material is presented in a friendly and easy to use manner which allows the student to pace their acquisition of knowledge and gain increasing confidence in order to succeed in understanding essential relevant concepts Other useful features of this book include starter diagnostic tests worked examples and self study tests with answers at the end of each unit In addition to Access or Foundation course students and their tutors to whom this book will prove essential it will have an appeal also as a revision text for those needing a refresher after a break in the subject In addition it will be of interest to members of the general public who wish to better educate themselves on chemical matters as it provides a clear and useful insight into areas such as health home chemicals business market trends and gardening **Chemistry**

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